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2020-21 X- PHYSICAL SCIENCE - (General Science - I) MODEL PAPER-3, E/M

	CLASS-X PART: A & B MAX, MARKS: 40 TIME: 1.35 Hrs			
	PART – A (Marks:30)			
	SECTIONS: I (3x2=6 Marks)			
1)	Which method can you used remove the impurities from its sulphide ores?			
2)	Why did the house-hold thing are burning due to overload explain it?			
3)	How can we get the virtual images in concave mirror, draw the ray diagram?			
4)	Why did the scientists fixed F-block elements in out of periodic table?			
5)	Explain the formation of images when we used concave and convex mirror in the rearview mirrors?			
6)	Why do we use flux and where can we use flux, give one example?			
SECTIONS II (2x4=8 Marks)				
7) 8) 9) 10	Why did we use Quantum Numbers Explain briefly? Why do we get eye defect. Write about types of eye defects? What type of Len we are using in microscope Explain with neat diagram? There is a fish inside the water. If that fish is exactly vertically bellow the bird the bird will appear to the fish to be?			
SECTIONS III (2x8=16 Marks)				
	1) Prove the ratio of V/I is a constant for a conductor? 2) Properties of elements and their trends in groups and in periods?			
13	3) How can you prove the whether given copper sulphate dry or wet substance?			
14	4) Write the Balancing chemical equation steps when propane react with Oxygen?			

PART - B (1x10=10 Marks)

BIT PAPER

1)	Which Pyrochemical process used when ore is heating abs	senc	y of air o
	Oxygen	[]
	(a) Roasting (b) Calcinations (c) blast (d) Furnace		
2)	To prevent the metal Corrosionsubstance is used		
	(a) Biopsil (b) Hydrogen (c) Phinoil (d) Ethanol		
3)	1 KWH = Jouls]
	a) $3.6 \times 10^7 \text{j}$ (b) $3.6 \times 10^5 \text{j}$ (c) $3.6 \times 10^6 \text{j}$ (d) $3.6 \times 10^8 \text{j}$	108j	
4)	What is the values of resistivity of insulators	[]
	(a) 10^{14} - $10^{16}\Omega$ -m (b) 10^{16} - $10^{18}\Omega$ -m		
	(c) 10^{12} - $10^{18}\Omega$ -m (d) 10^{18} - $10^{20}\Omega$ -m		
5)	What are the units for Atomic diameter	[]
	(a) m.m (b) c.m (c) K.m (d) P.m		
6)	Metal products are	[]
	(a) Chalcogens (b) Alkali metal (c) Helogen (d) Noble gas	es	
7)	Alkali metals we called as	[]
	(a) Base (b) metals (c) Acids (d) neutral		
8)	How many election are there in first orbital	[]
	(a) 8 (b) 18 (c) 2 (d) 0		
9)	Which element numbers is called as mandalivium	[]
	(a) 100 (b) 101 (c) 118 (d) 112		
10)	Value of flank constant []	
4	(a) 6.626×10^{-34} js (b) 6.636×10^{-34} js		
	(c) 6.262 X 10 ⁻³⁴ js (d) 6.616 X 10 ⁻³⁴ js		