## TET cum TRT - 2018

## TGT - GENERAL SCIENCE

1. The first Prime Minister of Independent India is
2. Jawarharlal Nehru
3. Indira Gandhi
4. Lal Bahadur Sastri
5. Gulzarilal Nanda
6. The gas used to extinguish fire is
7. Neon
8. Carbon dioxide
9. Carbon monoxide
10. Nitrogen
11. The Headquarters of HDFC Bank is located at
12. Mumbai
13. Bengaluru
14. Nainital
15. Delhi
16. SIM stands for
17. Self Identity Machine
18. Self Identity Module
19. Subscriber Identity Mode
20. Subscriber Identity Module
21. The Noble Prize 2017 in Literature was awarded to
22. Kazuo Ishiguro
23. Richard H. Thaler
24. Barry C Barish
25. Michael Rosbash
26. The present Governor of R.B.I is
27. Urgit Patel
28. Shakti Kanta Das
29. Ranga Rajan
30. Raghu Ram Rajan
31. World's women's Volley Ball championship 2018 was held at
32. Japan
33. Canada
34. India
35. Australia
36. The first Ebola Virus species was discovered in
37. 1956
38. 1966
39. 1976
40. 1986
41. CTRI is located at
42. Hyderabad
43. Guntur
44. Visakhapatnam
45. Rajahmundry
46. The first NTR National Award in 1996 was awarded to
47. N.T. Rama Rao
48. Akkineni Nageswara Rao
49. Amitabh Bachchan
50. M. Mohan Babu
51. The Cyclone that was affected in Andhra Pradesh in December 2018 was named as
52. Pethai
53. Titli
54. Hud Hud
55. Laila
56. The title given to Sri Potti Sree Ramulu was
57. Mahatma
58. Andhra Kesari
59. Amarajeevi
60. Andhra Tilak
61. The following person was not a recipient of Bharata Ratna Award in 1991
62. Rajiv Gandhi
63. Sardar Vallabhbhai Patel
64. Morarji Ranchhodji Desai
65. Gulzarilal Nanda
66. Rashtriya Madhyamika Shiksha Abhiyan (RMSA) was launched in
67. 2006
68. 2007
69. 2008
70. 2009
71. NIRF stands for
72. National International Ranking Framework
73. National Institutional Ranking Framework
74. National Integrated Ranking Framework
75. National Innovative Ranking Framework
76. The first Indian Prime Minister who visited Israel
77. Sri Man Mohan Singh
78. Sri Narendra Modi
79. Sri Atal Bihari Vajpayee
80. Sri Rajiv Gandhi
81. One of the following was called "All India war Memorial"
82. India Gate
83. Gate of India
84. Red Fort
85. Qutub Minar
86. The commonly used gas to fill incandescent light bulb is
87. Oxygen
88. Argon
89. Hydrogen
90. Neon
91. 'Beti Bachao Beti Padhao' is one of the flagship programmes of the Government launched in
92. 2015
93. 2014
94. 2016
95. 2017
96. The following scheme was launched by the Central Government for direct transfer of LPG subsidy all over the Country
97. PAHAL
98. NIFTEM
99. NMFP
100. RKSK
101. Which of these options shows the steps in the method of study laid down in Upanishads?
102. Sravana, Upanayana, Nidi- Dhyasana
103. Sravana, Upanayana, Manana
104. Sravana, Manana, Nidi- Dhyasana
105. Nidi- Dhyasana, Upanayana, Manana
106. This is true about the place of the teacher in Vedic education
107. To lead the student from the light of knowledge to the darkness of ignorance
108. Teachers were dishonoured by kings
109. The teacher did not essentially a man of character
110. The teacher was regarded as the spiritual and intellectual father for the students.
111. Identify the correct match
112. Register of pay bills- Account books
113. Memo book- Equipment register
114. Cumulative records- financial records
115. Contingent register- Educational Record
116. This is true about Private managements in establishing teacher education institutions
117. They have affiliating function.
118. They accord recognition to College of Education.
119. They have to get the recognition of the government on one hand, affiliation of the University on the other hand.
120. They conduct examinations and issue certificates at the under- graduate level of teacher education.
121. The aim of Adolescence Education Programme (AEP)
122. To provide interest- based, well organized complete vocational education to adolescents which will help them get a better job opportunity.
123. To ignite the mind of Adolescent youth to value their innate potential and nurture their creative ideas through hard work.
124. To empower young people with accurate, age appropriate and culturally relevant information, promote healthy attitudes and develop skills to enable them to respond to real life situations in positive and responsible ways.
125. To teach them the basics of entrepreneurship and educate them to take up small funding businesses in future.
126. When did the National Population Education Project initiated?
127. 1979
128. 1964
129. 1977
130. 1985
131. As per RTI Act 2005, which one is true regarding the time limit for filing the appeals?
132. First appeal should be filed within 30 days from the expiry of the prescribed time limit or from the receipt of the decision.
133. Second appeal should be filed within 90 days from the expiry of the prescribed time limit or from the receipt of the decision.
134. First appeal should be filed within 90 days of the date on which the decision was given or should have been given by the second appellate authority.
135. Second appeal should be filed within 30 days of the date on which the decision was given or should have been given by the first appellate authority.
136. When was the Universal Declaration of Human Rights (UDHR) adopted by the General Assembly in India?
137. $10^{\text {th }}$ December 1948
138. $10^{\text {th }}$ December 1950
139. $20^{\text {th }}$ August 1947
140. $\quad 26^{\text {th }}$ January 1948
141. Complete the sentence with appropriate word. According to NCF 2005, Secondary school is a period of intense physical change and formation of $\qquad$
142. ideas
143. identity
144. cognition
145. relations
146. Select the odd one out from the list of different streams given below.
147. the science stream
148. the commerce stream
149. the aesthetics stream
150. the arts stream
151. The number of intelligences in the Multiple Intelligences theory is
152. Three
153. Five
154. Nine
155. Seven
156. Performing simple arithmetic tasks promote
157. Verbal reasoning
158. Quantitative reasoning
159. Abstract reasoning
160. Short term memory
161. Encouraging students to come out with novel ideas will develop
162. Creativity
163. Intelligence
164. Emotional Intelligence
165. Self esteem
166. Reading and writing involve
167. Trial \& error learning
168. Cognitive learning
169. Psychomotor learning
170. Conditioning
171. A student learning to be punctual to the class is punctual to function
172. Imitation
173. Preparation
174. Transference
175. Transfer of learning
176. Assessment during the course of instruction rather than after it is completed
177. Summative Assessment
178. Continuous and Comprehensive Assessment
179. Pre Instructional Assessment
180. Formative Assessment
181. The general knowledge a student has about the world is
182. Episodic Memory
183. Short Term Memory
184. Semantic Memory
185. Implicit Memory
186. A state of fulfillment in which people realize their highest potential
187. Self efficacy
188. Self actualization
189. Deindividuation
190. Identification
191. Test in which an ambiguous stimulus is shown to a person and asked to describe it is
192. Projective test
193. Self report measure
194. Structured interview
195. Sentence completion test
196. A student to deal with his examination stress prepares a study schedule,
197. Adaptation
198. Problem focused coping
199. Assimilation
200. Emotion focused coping

## CONTENT

41. The length, breadth and depth of an underground water tank are 2 m , 1 m and 3 m respectively then the capacity of tank in litres is
42. 6
43. 60
44. 600
45. 6000
46. A car covers half the distance at a speed of 60 kmph and the other half at $100 / 9 \mathrm{~ms}^{-1}$, the average speed of the car is
47. 48 kmph
48. 50 kmph
49. $75 / 4 \mathrm{kmph}$
50. $320 / 9 \mathrm{kmph}$
51. If a force of 12 N produces an acceleration of $6 \mathrm{~ms}^{-2}$ on a mass $\mathrm{m}_{1}$ and $3 \mathrm{~ms}^{-2}$ on mass $\mathrm{m}_{2}$ then the acceleration produced by the same force on both the masses tied together is
52. $6 \mathrm{~ms}^{-2}$
53. $4 / 3 \mathrm{~ms}^{-2}$
54. $2 \mathrm{~ms}^{-2}$
55. 0
56. The position of the image when the object is placed in between F and C of a concave mirror
57. beyond C
58. at focal point
59. at centre of curvature
60. at pole
61. The characteristic property of sound that changes when we increase the loudness of sound of a T.V. is
62. speed
63. amplitude
64. frequency
65. wavelength
66. If 10 g of ice at $0^{\circ} \mathrm{C}$ is mixed with 10 g of water at $50^{\circ} \mathrm{C}$ then the resultant temperature of the mixture is
67. $25^{\circ} \mathrm{C}$
68. $100^{\circ} \mathrm{C}$
69. $50^{\circ} \mathrm{C}$
70. $\quad 0^{\circ} \mathrm{C}$
71. A uniform wire of resistance $10 \Omega$ is melted and recast into a wire of length three times that of the original, then the resistance of the wire newly formed is
72. $30 \Omega$
73. $90 \Omega$
74. $10 \Omega$
75. $10 / 3 \Omega$
76. A charged particle ' $q$ ' is moving with a speed ' $v$ ' perpendicular to the magnetic field of induction ' $B$ ', its path is
77. straight line
78. elliptical
79. circular
80. triangular
81. If a body contains ' $n$ ' number of positive and ' $n$ ' number of negative charges, on rubbing it looses $x$ charges. The number of positive and negative charges present on the body are
82. $n$ positive, $(n-x)$ negative charges
83. $(\mathrm{n}-x)$ positive, $(\mathrm{n}-x)$ negative charges
84. $(\mathrm{n}+x)$ positive, $(\mathrm{n}+x)$ negative charges
85. $(\mathrm{n}-x)$ positive, n negative charges
86. 'Uttarayanam' means
87. Moon looks like travelling towards north of the Sky.
88. Earth looks like travelling towards north of the Sky
89. Sun looks like travelling towards north of the Sky
90. Pole star looks like travelling towards north of the Sky
91. Length of two bodies are measured as $l_{l}=(20 \pm 0.5) \mathrm{m}$ and $l_{2}=(50 \pm 0.5) \mathrm{m}$. The difference in length and error in the measurement is
92. $(\mathbf{3 0} \pm \mathbf{1}) \mathrm{m}$
93. $(30 \pm 0) \mathrm{m}$
94. $(30 \pm 0.25) \mathrm{m}$
95. $(30 \pm 0.10) \mathrm{m}$
96. If the vertical component of a vector is equal to its horizontal component then the angle made by the vector with x -axis is
97. $\mathbf{4 5}^{\circ}$
98. $30^{\circ}$
99. $0^{\circ}$
100. $90^{\circ}$
101. The power of the machine gun is 2 KW , it fires 300 bullets per minute of mass 5 g each then the velocity of each bullet is
102. $600 \mathrm{~ms}^{-1}$
103. $400 \mathrm{~ms}^{-1}$
104. $500 \mathrm{~ms}^{-1}$
105. $300 \mathrm{~ms}^{-1}$
106. The fringe separation in Young's double slit experiment when the slits are separated by 0.28 mm and the screen is placed 1.4 m away (the light of wavelength 600 mm is used) is
107. $\quad 1.2 \times 10^{-10} \mathrm{~m}$
108. 3 cm
109. $\mathbf{3} \mathbf{~ m m}$
110. 0.3 mm
111. A tuning fork vibrates at 350 Hz and the speed of sound in air is 350 $\mathrm{ms}^{-1}$. The length of the shortest organ closed pipe that will resonate with the tuning fork is
112. 50 cm
113. 1 cm
114. 100 cm
115. 25 cm
116. A body cools down from $80^{\circ} \mathrm{C}$ to $50^{\circ} \mathrm{C}$ in 5 minutes. The time taken by the same body to cool down from $60^{\circ} \mathrm{C}$ to $30^{\circ} \mathrm{C}$ is (room temperature $=20^{\circ} \mathrm{C}$ )
117. 5 minutes
118. less than 5 minutes
119. depends on mass of the substance
120. more than 5 minutes
121. A 200 pf capacitor is charged by a 100 V supply. It is then disconnected from the supply and is connected to another uncharged 200 pf capacitor. The electrostatic energy lost in this process is
122. $2 \times 10^{-6} \mathrm{~J}$
123. $0.5 \times 10^{6} \mathrm{~J}$
124. $0.5 \times 10^{-6} \mathrm{~J}$
125. $2 \times 10^{6} \mathrm{~J}$
126. An LCR series circuit with $\mathrm{L}=100 \mathrm{mH}, \mathrm{C}=100 \mu \mathrm{~F}, \mathrm{R}=120 \Omega$ is connected to an AC source. Resonating frequency of the circuit is close to
127. 100 Hz
128. 50 Hz
129. 0 Hz
130. 311 Hz
131. The angular momentum of an electron in the hydrogen atom is $\frac{3 h}{2 \pi}$ .Here $h$ is Plank's constant. The kinetic energy of this electron is
132. 4.35 eV
133. $\quad 1.51 \mathrm{eV}$
134. 3.4 eV
135. 6.8 eV
136. When a base current in a Transistor is changed from $30 \mu \mathrm{~A}$ to $80 \mu \mathrm{~A}$, the collector current changes from 2.0 mA to 4.5 mA then the current gain is
137. 0.02
138. 20
139. 50
140. 5
141. The following condition will increase the evaporation of water
142. Less exposed surface area of water
143. Increase in temperature of water
144. Decrease in temperature of water
145. Adding impurities to water
146. Acid rains are the combination of
147. carbonic acid +nitric acid + rain water
148. carbonic acid + acetic acid + rain water
149. acetic acid + sulphuric acid + rain water
150. nitric acid + acetic acid + rain water
151. The incorrect statement regarding "burning test of fibers"
152. Silk gives smell like burning hair
153. Cotton gives smell like burning paper
154. Nylon gives smell like burning hair
155. Acrylic melts in the flame and shrinks
156. The figure $\mathrm{A}, \mathrm{B}, \mathrm{C}, \mathrm{D}$ represents the layers of earth. Which among the following is correct

157. A - Earth's crust, B - Water, C - Petroleum, D - Natural gas
158. A - Earth's crust, B - Petroleum, C - Water, D - Natural gas
159. A - Earth's crust, B - Water, C - Natural gas, D - Petroleum
160. A - Earth's crust, B - Natural gas, C - Petroleum, D Water
161. The colourless, odourless, combustible and a renewable source of energy is
162. Nitrogen gas
163. Carbon dioxide gas
164. Carbon monoxide gas

## 4. Hydrogen gas

66. On converting $35^{\circ} \mathrm{C}, 73^{\circ} \mathrm{C}$ and $85^{\circ} \mathrm{C}$ in to Kelvin scale, the correct sequence of temperatures will be
67. $308 \mathrm{~K}, 346 \mathrm{~K}, 368 \mathrm{~K}$
68. $308 \mathrm{~K}, 336 \mathrm{~K}, 358 \mathrm{~K}$
69. 308 K, 346 K, 358 K
70. $308 \mathrm{~K}, 356 \mathrm{~K}, 368 \mathrm{~K}$
71. Identify the pair which is not Isotopes.
72. $\quad{ }_{6}^{12} \mathrm{X},{ }_{6}^{14} \mathrm{Y}$
73. $\quad{ }_{17}^{35} \mathrm{X},{ }_{17}^{37} \mathrm{Y}$
74. ${ }_{15}^{30} \mathrm{X},{ }_{15}^{31} \mathrm{Y}$
75. ${ }_{7}^{14} \mathrm{X},{ }_{6}^{13} \mathrm{Y}$
76. In the following sequences of atomic numbers, the sequence that represents 'representative elements'
77. $3,20,13,17$
78. $11,22,15,10$
79. $19,11,16,18$
80. $14,10,3,18$
81. Valence shell electron pair repulsion theory mainly failed to explain
82. Bond angles of the molecules
83. Electron pair repulsion
84. Strength of the bonds
85. Lone pair - bond pair repulsion
86. The IUPAC name of

87. 2, 5, 5-trimethyl hexane
88. 2, 2, 5-trimethyl hexane
89. 2-methyl, 5, 5-dimethyl hexane
90. 5,5-dimethyl, 2-methyl hexane
91. The Kinetic energy of 5 moles of nitrogen gas at $27^{\circ} \mathrm{C}$ is $\left[\mathrm{R}=8.314 \mathrm{Joul} \mathrm{K}^{-1}\right]$
92. $\quad 18.706 \mathrm{~J}$
93. 62382 J
94. $\quad 1647.1 \mathrm{~J}$
95. $\quad \mathbf{1 8 7 0 6 . 5}$ J
96. The correct statement about aqueous solution of an acid and a base
97. Higher the $\mathrm{p}^{\mathrm{H}}$, strong acid; higher the $\mathrm{p}^{\mathrm{H}}$, weak base
98. Lower the $\mathrm{p}^{\mathrm{H}}$, strong acid; lower the $\mathrm{p}^{\mathrm{H}}$, strong base
99. Lower the $\mathrm{p}^{\mathrm{H}}$, weak acid; lower the $\mathrm{p}^{\mathrm{H}}$, strong base
100. Lower the $\mathbf{p}^{\mathrm{H}}$, strong acid; lower the $\mathbf{p}^{\mathrm{H}}$, weak base
101. The correct order of increasing ionic character of hydrides of alkali metals is
102. $\mathrm{NaH}>\mathrm{CsH}>\mathrm{RbH}>\mathrm{LiH}>\mathrm{KH}$
103. $\mathrm{RbH}>\mathrm{CsH}>\mathrm{NaH}>\mathrm{KH}>\mathrm{LiH}$
104. $\mathbf{L i H}<\mathbf{N a H}<\mathbf{K H}<\mathbf{R b H}<\mathbf{C s H}$
105. $\mathrm{LiH}>\mathrm{NaH}>\mathrm{CsH}>\mathrm{KH}>\mathrm{RbH}$
106. In the presence of reducing agents like lithium aluminum hydride or diborane, carboxylic acid $\mathrm{R}-\mathrm{COOH}$ gives
107. $\mathrm{R}-\mathrm{CHO}$
108. $\quad{ }_{R}^{R}>C=O$
109. $\mathrm{R}-\mathrm{O}-\mathrm{R}$
110. $\mathrm{R}-\mathrm{CH}_{2} \mathrm{OH}$
111. This is produced on heating of tetra fluoro ethane with free radical or per sulphate catalyst at high pressure,
112. Bakelite
113. Melamine
114. Teflon
115. Nylon 6.6
116. The only covalent halide of alkaline earth metal which is soluble in organic solvents is
117. $\mathrm{MgCl}_{2}$
118. $\mathrm{CaCl}_{2}$
119. $\mathbf{B e C l}_{2}$
120. $\mathrm{BaCl}_{2}$
121. The tendency of catenation in group-14 elements follows the order
122. $\mathrm{C}>\mathrm{Si}>\mathrm{Sn}>\mathrm{Ge}$
123. $\mathbf{C} \gg \mathbf{S i}>\mathbf{G e} \approx \mathbf{S n}$
124. $\mathrm{Si}>\mathrm{C}>\mathrm{Sn}>\mathrm{Ge}$
125. $\mathrm{Ge}>\mathrm{Sn}>\mathrm{Si}>\mathrm{C}$
126. The consequence of Global Warming is
127. Increase in average temperature of the earth
128. Formation of Himalayan Glaciers
129. Increased biochemical oxygen demand
130. Increased chemical oxygen demand
131. In saponification the glyceryl esters of fatty acids are treated with
132. Aqueous magnesium hydroxide
133. Aqueous sodium hydroxide
134. Aqueous lithium hydroxide
135. Ethanol solution
136. The pair of ions having same electronic configuration is
137. $\mathrm{Sc}^{3+}, \mathrm{Cr}^{3+}$
138. $\mathbf{F e}^{3+}, \mathbf{M n}^{2+}$
139. $\mathrm{Fe}^{3+}, \mathrm{Co}^{3+}$
140. $\mathrm{Cr}^{3+}, \mathrm{Fe}^{3+}$
141. The fact aboutLactobacillus and Thiomargaritanamibiensis respectively is
142. Both are available in human digestive tract.
143. It absorbs nutrients by way of osmosis - it swallows food material
144. It can be seen through microscope - it can be seen unaided eye
145. It turns milk into curd - it turns curd into milk
146. The characteristic of E-Horizon is
147. It is found between B-Horizon and C-Horizon
148. Allows dripping of minerals and water through it to next horizons
149. It contains clay and mineral deposits
150. It is made up of humus
151. Antibiotics can kill
152. Virus only
153. Bacteria only
154. Virus and Bacteria
155. All micro organisms
156. The ascending sequence of hirarchial order in systamatics is
157. Species - genus - order - family - class - phylum - kingdom
158. Species - genus - class - family - order - phylum - kingdom
159. Species - genus - family - class - order - phylum - kingdom
160. Species - genus - family - order - class - phylum kingdom
161. When anti-'A' and anti-'B' sera are added to two blood samples of two persons, agglutination is appeared. These samples respectively belong to
162. Blood group - 'AB'
163. Blood group - 'B', 'A'
164. Blood group - 'O'
165. Blood group - 'A', 'B'
166. Collenchyma are characterised by the presence of
167. isodiamatric cell wall with thickening all over the cell wall
168. soft, thin wall with large air cavities
169. elongated (longer) cells with thick wall
170. elongated (longer) cells with thin wall
171. Identify the correct statements.
A. Plants with tap root system have leaves with reticulate venation
B. Plants with fibrous root system have leaves with reticulate venation
C. Plants with tap root system have leaves with parallel venation
D. Plants with fibrous root system have leaves with parallel venation
172. $\mathrm{A}, \mathrm{D}$ and C
173. D and A
174. B and C
175. C, B and D
176. A student saw 20-5-10 on a fertilizer bag. The numbers on the bag represent respectively are
177. Phosphorus - Nitrogen - Potassium
178. Nitrogen - Zinc - Potassium
179. Nitrogen - Potassium - Super Phosphorus
180. Nitrogen - Phosphorus - Potassium
181. Identify the incorrect one

## Excretion

1. same in nature
same in nature
2. active in nature - passive in nature
3. involves in movements - involves in movements of materials
of materials
4. Ex: Tears

Ex: resins
90. In the process of inhalation in human beings

1. the internal pressure of lungs decreases
2. the internal pressure of lungs increases
3. the volume of the chest cavity decreases
4. the ribs stop the expansion of lungs
5. The organs that donot carryout digestion in human beings
6. Large intestine, small intestine, rectum
7. Oesophagus, large intestine, rectum
8. Buccal cavity, oesophagus, large intestine
9. Oesophagus, Buccal cavity, rectum
10. The term plants was used by these scientists in their classification of living beings.
11. Linnaeus, Chatton
12. Haeckel, Chatton
13. Woese et al, Cavilier-smith
14. Copeland, Whittaker
15. The correct statement about G1 phase
16. Cell is metabolically inactive
17. DNA in the cell does not replicate
18. It is not a phase of synthesis of macromolecules
19. Cell stops growing

94 The symptoms of Folic acid deficiency are

1. Scaly skin, Paralysis
2. Walking problem, Fracture of bones
3. Dermatitis, loss of memory
4. Loss of leucocytes, intestinal mucus problems
5. In human beings the blood capillaries do not contain this character.
6. These are microscopic double layered vessels
7. The leucocytes can squeeze out of the capillary wall
8. They allow diffusion of various substances
9. They establish continuity between arteries and veins
10. The correct order of dominant and recessive traits of pea plant selected by Mendel respectively
11. Flower colour : white - purple
12. Pod shape : constricted - inflated
13. Flower position : axial - terminal
14. Pod colour : yellow - green
15. To consider a soil as an organic soil the minimum percentage of organic matter it should contain is
16. $20 \%$
17. $50 \%$
18. $\mathbf{3 0 \%}$
19. $60 \%$
20. One of the following is not a characteristic of desert ecosystem.
21. Decomposers are few
22. Succulent cacti are present
23. Thermophilic fungi and bacteria are present
24. Roots are not much branched and not extensive
25. Regarding the process of cloning of Dolly the correct statement is
26. A cell from mammary gland of Scottish black face ewe was used
27. An egg of Finn Dorset ewe was used
28. Dolly was given birth by Finn Dorset ewe
29. Dolly was given birth by Scottish black face ewe
30. Flamingos can be seen in these sanctuaries
31. Krishna and Kinnerasani
32. Pranahitha and Kawal
33. Koringa and Kolleru
34. Pakala and Eturunagaram
35. Carpagonium is the female sex organ of
36. Ectocarpus
37. Polysiphonia
38. Laminaria
39. Fucus
40. Pick the incorrect statement.
41. The mycelium of 'algal fungi' is aseptate and multinucleate
42. The sexual spores of 'sac fungi' are called conidiospores
43. The sex organs are absent in 'bracket fungi'
44. The 'imperfect fungi' reproduce only by means of asexual or vegetative structures.
45. Unisexual flowers are seen in
46. Crotalaria and Carrot
47. Achyranthus and Oryza
48. Colocasia and Ficus
49. Solanum and Cassia
50. The endosperm cell of a plant has 24 chromosomes. The number of bivalents formed in pachytene stage of meiosis-I are
51. 8
52. 16
53. 24
54. 48
55. The location of the NADP reductase in thylakoid membrane of chloroplast is towards
56. Stromal side of PS II
57. Stromal side of PS I
58. Lumen side of PS II
59. Lumen side of PS I
60. Inheritance of flower colour in snapdragon is an example for
61. Pseudo dominance
62. over dominance
63. Codominance
64. Incomplete dominance
65. Assertion : Sulphur deficiency symptoms first appear in the immature organs.
Reason : Sulphur is highly mobile in plants.
66. Both $A$ and $R$ are true. $R$ is correct explanation for $A$
67. Both A and R are true. R is not correct explanation for A
68. $\quad A$ is true. $R$ is false
69. A is false. R is true
70. Match list I with list II and select the correct answer using the codes given below the lists.

List-I
A) Transferase
B) Hydrolase
C) Lyase
D) Ligase

A $\quad$ B $\quad$ C

1. II I III IV
2. III II I IV
3. III I IV II
4. I II IV III
5. Identify the incorrect sequence of steps in polymer chain reaction (PCR) cycle.
6. Annealing $\rightarrow$ extension $\rightarrow$ denaturation
7. Denaturation $\rightarrow$ annealing $\rightarrow$ extension
8. Extension $\rightarrow$ denaturation $\rightarrow$ annealing
9. Annealing $\rightarrow$ denaturation $\rightarrow$ extension
10. The hybrid plant 'Pomato' was developed through
11. Genetic engineering
12. Hybridisation
13. Somatic hybridisation
14. Mutation breeding
15. Assertion (A) : Species is an ecological unit

Reason (R) : It shares same ecological niche

1. Both $A$ and $R$ are true. $R$ is correct explanation for $A$
2. Both A and R are true. R is not correct explanation for A
3. A is true. R is false
4. A is false. R is true.
5. Pick up the secondary lymphoid organ from the following.
6. Bursa fabricius of birds
7. Bone marrow of mammals
8. Thymus gland of mammals
9. Spleen of mammals
10. In nervous tissue, the supporting cells developed from mesoderm are
11. Astrocytes
12. Microglial cells
13. Ependymal cells
14. Satellite cells
15. The digestive juice with steapsin
16. Gastric juice
17. Bile juice
18. Pancreatic juice
19. Intestinal juice
20. The causative organism of the disease ring worm is
21. Salmonella
22. Streptococcus
23. Rhinovirus
24. Microsporum
25. Match the following

|  | List -1 |  | List -2 |
| :--- | :--- | :--- | :--- |
| A. | Fossa ovalis | I. | Pulmonary arch |
| B. | Bundle of His | II. | Post caval vein |
| C. | Eustachian valve | III. | Inter atrial septum |
| D. | Semilunar valves | IV. | Inter ventricular septum |
|  |  | V. | Atrio ventricular septum |


|  | A | B | C | D |
| :--- | :--- | :--- | :--- | :--- |
| 1. | I | II | V | III |
| 2. | II | V | I | IV |
| 3. | IV | I | III | II |
| 4. | III | IV | II | I |

117. Study the following and pick up the mismatched pair.
118. Gonorrhoea - Neisseria
119. Syphilis - Treponema
120. AIDS - Human Immunodeficiency Virus
121. Cervix cancer - Trichomonas
122. Karyotype of Kleinfelter's syndrome is
123. $\mathrm{AA}+\mathrm{XO}$
124. $\mathbf{A A}+\mathbf{X X Y}$
125. $A A A+X X$
126. $\mathrm{AA}+\mathrm{XY}$
127. The gases responsible for green house effect are
128. Methane, Carbon dioxide
129. Carbon dioxide, Ozone
130. Carbon dioxide, Sulphur dioxide
131. Carbon monoxide, Methane
132. Study the following statements and pick up the correct one
133. Carcinoma is the cancer of connective tissues
134. Leukemia is the cancer of epithelial tissues
135. Sarcoma is the cancer of connective tissues
136. Lymphoma is the cancer of nervous tissue

## METHODOLOGY

## 121. Science discovers knowledge

1. From ancient texts or revelations of inspired individuals.
2. From personal insight, or insight of others
3. From evidence generated by observation or by experimentation.
4. From personal feelings and perceptions
5. This is an example for scientific hypothesis
6. It's bright and hot outside.
7. When the sun is out, it tends to make it bright outside
8. It's bright outside because it is day time.
9. The temperature outside today is $32^{\circ} \mathrm{C}$
10. Writing the similarities and differences between a copper wire and a glass slab is an example of
11. Observing
12. Hypothesizing
13. Communicating
14. Inferring
15. The odd one out of the following is
16. Skepticism
17. Inferring
18. Open mindedness
19. Respecting others opinion
20. "Principia" and "Opticks" are pioneering works of
21. Copernicus
22. Newton
23. Galileo
24. Einstein
25. The ancient Indian scholar who calculated the exact value of pi
26. Varahamitra
27. Bhaskaracharya
28. Aryabhatta
29. Brahmagupta
30. The concept that Einstein banished from Physics
31. Photo electricity
32. Relativity
33. Electromagnetic Induction
34. Ether
35. The person who awarded the Nobel Prize in recognition of discovery of the elements Radium and Polonium is
36. Marie Curie
37. Arrhenius
38. Rutherford
39. Ramsay
40. A teacher comparing image formation in Camera with Human Eye is an example of
41. Correlation of Chemistry and biology
42. Correlation of Physics and Chemistry
43. Correlation of Physics and Biology
44. Correlation of Physics and Art
45. A person who engages in critical thinking, research, and reflection possesses this value
46. Aesthetic value
47. Moral value
48. Intellectual value
49. Cultural value
50. This is an aim of science teaching
51. do not develop critical observation and thought
52. provides training to student in scientific method
53. more useful to the teacher
54. will not prepare students for a vocation in science
55. This is a specific behaviour of affective domain
56. Understands nature of science
57. Appreciates beauty of science
58. Applies principles of science
59. Sets up an experiment of science
60. This is an example of manipulation skill
61. Explaining the formation of stars
62. Arranging apparatus for preparation of oxygen
63. Comparing lunar and solar eclipse
64. Defining light year
65. The hierarchical structure of Cognitive domain is
66. Evaluation, Knowledge, Synthesis, Application
67. Application, Synthesis, Evaluation ,Knowledge,
68. Synthesis, Knowledge, Application, Evaluation
69. Knowledge, Application, Synthesis, Evaluation
70. A student who uses kinetic molecular theory to explain rates of reactions is at this level of cognitive domain
71. Evaluation
72. Analysis
73. Synthesis
74. Application
75. This method gives more emphasis on teacher presentation and has no place for the participation of the students
76. Project method
77. Lecture method
78. Heuristic method
79. Laboratory method
80. A teacher who begins the lesson narrating about Newton and the apples falling down is using this method
81. Demonstration method
82. Lecture - Demonstration method
83. Historical method
84. Project method
85. The principle of teaching through deductive method is
86. From general to particular
87. From specific to general
88. From illustrations to law
89. From concrete to abstract
90. This is a pre instruction micro skill
91. stimulus variation
92. closure
93. reinforcement
94. set induction
95. Making different Optical instruments like Microscope and Telescope is
96. Drill type project
97. Problem type project
98. Producer type project
99. Consumer type project
100. This is not considered by a science teacher while preparing a year plan
101. Number of units in syllabus
102. Number of science periods per week
103. Number of learning activities to be included
104. Number of working days per year per term
105. The step which precedes Presentation in Herbartian lesson plan is
106. Recapitulation
107. Introduction
108. Comparison
109. Generalisation
110. This topic will be included in a unit plan on 'Motion'
111. Laws of Motion
112. Units of Measurement
113. Collisions and Impulse
114. Simple Harmonic Motion
115. This is an example for contrived experience
116. Finding mass of an object
117. Planetarium
118. Role play
119. Motion pictures
120. The most abstract experience of the following
121. Recordings
122. Demonstration
123. Verbal symbols
124. Visual symbols
125. Out of the following this is a projection device
126. DVD
127. Hard disc
128. Microphone
129. OHP
130. Chart is an example for
131. Audio visual aid
132. Audio aid
133. Activity aid
134. Graphic aid
135. This is included in Good laboratory work practices
136. smelling and tasting chemicals
137. not washing hands before and after lab
138. not confining to long hair and loose clothing
139. using damaged equipment and glassware
140. The Register in Science lab which has details of glassware - funnels, beakers, burettes etc., only is
141. Permanent stock register
142. Breakable stock register
143. Consumable stock register
144. Order register
145. The chemical substance that has to be kept in water in the Science lab is
146. Potassium
147. Iodine
148. Phosphorus
149. Carbon di sulphide
150. If an acid is splashed on the skin accidentally in the lab, it should be washed at once with plenty of water and then with.....
151. soap.
152. oil.
153. weak base.
154. lime water.
155. Repetition of Newton's Laws of motion at same level of difficulty in classes $7^{\text {th }}$ and $8^{\text {th }}$ lacks the quality of:
156. Horizontal organization
157. Vertical organization
158. Relevancy
159. Flexibility
160. Knowledge is compartmentalized in:
161. Learner centred curriculum
162. Activity centred curriculum
163. Subject centred curriculum
164. Experience centred curriculum
165. This is a Science museum being run by an NGO
166. Centre for Cellular and Molecular Biology, Hyderabad
167. Visvesvaraya Industrial \& Technological Museum, Bengaluru
168. Council of Scientific and Industrial Research, New Delhi
169. B.M. Birla Science Centre, Hyderabad
170. This is not a characteristic of Science Exhibition
171. Students think and work like a scientist
172. Original concept, publishable
173. Planning and designing an experiment is required
174. Inquiry, question, and problem are not part of this activity
175. The type of question which would be the best to use to introduce a new topic is
176. A low order question
177. A close-ended question
178. A question that calls for convergent thinking
179. A question that calls for divergent thinking
180. Questioning in the class-room

## 1. Clarifies the subject-matter

2. Develops inactivity
3. Is a wastage of time
4. Creates indiscipline
5. This could be most effectively documented by an anecdotal record
6. Improvement in drawing skills
7. Progress in computational skills
8. Problem solving ability
9. Problem of adjustment in project work
10. An evaluation that is generally carried out at the end of a course to assign students a course grade is called
11. Diagnostic evaluation
12. Formative evaluation
13. Placement evaluation
14. Summative evaluation
15. The objective which could be properly assessed by performance assessment
16. Student will interpret the periodic table
17. Student will find out the width of a glass slab
18. Student will distinguish facts from scientific facts
19. Student will define atomic number
