Question Paper Preview

Question Paper Name:Chemical EngineeringSubject Name:Chemical Engineering

Mathematics

Number of Questions:50Display Number Panel:YesGroup All Questions:No

Question Number: 1 Question Id: 67809417024 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

If
$$A = \begin{pmatrix} 2 & -1 & 0 \\ 3 & 4 & 7 \end{pmatrix}$$
 and $B = \begin{pmatrix} 5 & 2 & -3 \\ 1 & 0 & -2 \end{pmatrix}$ then $2A+3B =$

Options:

$$\begin{pmatrix} 19 & 4 & -9 \\ 9 & 8 & 8 \end{pmatrix}$$

$$\begin{pmatrix} -19 & -4 & 9 \\ 9 & 8 & -8 \end{pmatrix}$$

$$\binom{18}{9} \binom{4}{8} \binom{-9}{8}$$

$$\begin{pmatrix} 17 & 5 & -9 \\ 8 & 8 & 9 \end{pmatrix}$$

Question Number : 2 Question Id : 67809417025 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

If
$$A = \begin{pmatrix} 2 & -3 & 0 \\ 1 & 4 & -1 \end{pmatrix}$$
 and $B = \begin{pmatrix} 6 & 1 \\ 3 & 0 \\ 5 & 2 \end{pmatrix}$ then $(AB)^T =$

$$A^{T}B^{T}$$

$$_{2}$$
 $B^{T}A^{T}$

$$_{3}$$
 (BA)^T

$$_{4.}$$
 AB^{T}

Question Number : 3 Question Id : 67809417026 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

If two rows or two columns of a determinant are identical then the value of the determinant is

Options:

- 1. 2
- 2. -1
- 3. 0
- 4. -2

Question Number: 4 Question Id: 67809417027 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Options:

- , -1
- 2 0
- 3. 1
- 4. 2

Question Number : 5 Question Id : 67809417028 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The adjoint of the square matrix
$$A = \begin{pmatrix} 2 & 5 & 1 \\ 3 & 1 & 2 \\ 4 & 3 & 1 \end{pmatrix}$$
 is

$$\begin{pmatrix} -5 & -2 & 9 \\ 5 & -2 & -1 \\ 5 & 14 & -13 \end{pmatrix}$$

$$\begin{pmatrix} 5 & 2 & 9 \\ 5 & -2 & -1 \\ 5 & 14 & -13 \end{pmatrix}$$

$$\begin{pmatrix} -5 & -2 & 9 \\ -5 & -2 & -1 \\ -5 & 14 & -13 \end{pmatrix}$$

$$\begin{pmatrix} -5 & -2 & -9 \\ 5 & 2 & 1 \\ 5 & 14 & -13 \end{pmatrix}$$

Question Number : 6 Question Id : 67809417029 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Resolve into partial fractions: $\frac{5}{(2x-1)(3x-1)}$ =

Options:

$$\frac{8}{2x-1} + \frac{5}{3x-1}$$

$$\frac{10}{2x-1} - \frac{15}{3x-1}$$

$$\frac{11}{3x-1} + \frac{7}{2x-1}$$

$$\frac{1}{2x-1} + \frac{2}{3x-1}$$

Question Number: 7 Question Id: 67809417030 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Resolve into partial fractions: $\frac{3x-1}{(x-1)(x-2)(x-3)} =$

$$\int_{1}^{2} \frac{2}{x-1} + \frac{5}{x-2} - \frac{4}{x-3}$$

$$\frac{-1}{x-1} + \frac{5}{x-2} - \frac{4}{x-3}$$

$$\frac{1}{x-1} + \frac{5}{x-2} + \frac{4}{x-3}$$

$$\frac{1}{4} \cdot \frac{1}{x-1} - \frac{5}{x-2} + \frac{4}{x-3}$$

Question Number: 8 Question Id: 67809417031 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

If $tan A = \frac{1}{2}$ and $tan B = \frac{1}{3}$ then tan (A - B) =

Options:

- 1. 7
- $\frac{-1}{7}$
- 2 5
- 1 3

Question Number: 9 Question Id: 67809417032 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\cot 2A + \tan A =$

Options:

- 1 sin2A
- 2. cos2A
- 3. sec2A
- 4. cosec2A

Question Number: 10 Question Id: 67809417033 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\frac{1-\cos 2A+\sin 2A}{1+\cos 2A+\sin 2A} =$

Options:

- 1. sinA
- 2. cosA
- 3 tanA
- 4. cotA

Question Number: 11 Question Id: 67809417034 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\sin \frac{\pi}{5} \sin \frac{2\pi}{5} \sin \frac{3\pi}{5} \sin \frac{4\pi}{5} =$

Options:

- 1. 15
- 2 16
- -5 3 16
- 4 15

Question Number: 12 Question Id: 67809417035 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\cos 20^{\circ} + \cos 100^{\circ} + \cos 140^{\circ} =$

- 1. 0
- 2. 3
- 3. 1
- 4. -3

Question Number: 13 Question Id: 67809417036 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\sum a(b^2 + c^2)\cos A$ is

Options:

- 1. 2abc
- 2. 4abc
- 3. 3abc
- 4. 5abc

Question Number: 14 Question Id: 67809417037 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $(a-b)^2 cos^2 \left(\frac{c}{2}\right) + (a+b)^2 sin^2 \left(\frac{c}{2}\right)$ is

Options:

- $_1$ C^3
- 2. C
- 3 C5
- 4. C2

Question Number: 15 Question Id: 67809417038 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $2tan^{-1}\left(\frac{1}{3}\right) + tan^{-1}\left(\frac{1}{7}\right)$ is

Options:

- $1 \pi/4$
- $_{2}$ $\pi/2$
- $_{3.} \pi/6$
- 4. $\pi/3$

Question Number: 16 Question Id: 67809417039 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The general solution of $4\cos^2 x - 3 = 0$ is

Options:

$$2n\pi \pm \frac{\pi}{6}$$

$$_{2}$$
 $2n\pi \pm \frac{7\pi}{6}$

$$3n\pi \pm \frac{5\pi}{6}$$

$$2n\pi \pm \frac{11\pi}{6}$$

Question Number: 17 Question Id: 67809417040 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

If $tan^{-1}x + tan^{-1}y + tan^{-1}z = \frac{\pi}{2}$, then the value of xy + yz + zx is

Options:

- 1. -1
- 2. 3
- 3. 5
- 4. 1

Question Number: 18 Question Id: 67809417041 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The modulus of a complex number $\sqrt{3} + i$ is

- 1. -2
- 2. 3
- 3. 2
- 4. 5

If $x + \frac{1}{x} = 2\cos\theta$ then the value of $x^n + \frac{1}{x^n}$ is

Options:

- $1 \cos n\theta$
- $_2$ -2 cos $n\theta$
- $_{3} 3\cos\theta$
- $\frac{2\sin n\theta}{\theta}$

Question Number: 20 Question Id: 67809417043 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The centre of the circle: $x^2 + y^2 - 2x + 6y - 6 = 0$ is

Options:

- $_{1.}$ (1,3)
- $_{2}$ (2,3)
- $_{3.}$ (1,-3)
- 4 (-1,3)

Question Number: 21 Question Id: 67809417044 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The radius of the circle: $5x^2 + 5y^2 - 6x + 8y - 75 = 0$ is

Options:

- 1. -4
- 2. 4
- 3. 2
- 4 3

Question Number: 22 Question Id: 67809417045 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The equation of the parabola with vertex (2, -1) and focus (2, -3) is

$$x^2 - 4x + 8y + 12 = 0$$

$$x^2 - 4x - 8y - 12 = 0$$

$$x^2 + 4x - 8y - 12 = 0$$

$$x^2 + 5x - 8y - 11 = 0$$

Question Number: 23 Question Id: 67809417046 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The centre of the ellipse: $9x^2 + 25y^2 - 18x + 100y - 116 = 0$ is

Options:

$$_{1}$$
 (2,-1)

$$_{2}$$
 $(-1,-2)$

$$_{3}$$
 (1,-2)

Question Number: 24 Question Id: 67809417047 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The focus of the hyperbola: $\frac{x^2}{25} - \frac{y^2}{144} = 1$ is

Options:

$$(-13,0)$$

$$_{3}$$
 (13, -1)

Question Number: 25 Question Id: 67809417048 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The length of the major axis of the ellipse: $4x^2 + 3y^2 = 48$ is

- 1.10
- 2. 11
- 3. 8
- 4. 13

Question Number: 26 Question Id: 67809417049 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\lim_{x\to 1} \frac{x^3-1}{x-1}$ is

Options:

- 1. 3
- 2. -3
- 3. 2
- 4. 1

Question Number: 27 Question Id: 67809417050 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

If $y = \frac{a+bx}{b-ax}$ then the derivative of y with respect to x is

$$\int_{1}^{a^2+b^2} \frac{a^2+b^2}{(b-ax)^2}$$

$$\frac{a^2+b^2}{(b+ax)^2}$$

$$\frac{a^2-b^2}{(b-ax)^2}$$

$$\frac{a+b}{(b-ax)^2}$$

If
$$y = x^3 e^x$$
 then $\frac{dy}{dx}$ is

$$(x-3)x^2e^x$$

$$(x-2)x^3e^x$$

$$\int_{3.} (x+3)x^2 e^x$$

$$(x-1)x^3e^x$$

Question Number : 29 Question Id : 67809417052 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

If $y = \sec x + \tan x$ then $\frac{dy}{dx}$ is

Options:

- $\int_{1}^{\infty} y \cos x$
- $_2$ y sec x
- $y = -y \sin x$
- $y \tan x$

Question Number : 30 Question Id : 67809417053 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

If $y = \frac{2+3 \sinh x}{3+2 \sinh x}$ then the derivative of y with respect to x is

$$\int_{1}^{\infty} \frac{5\cosh x}{(3+2\sinh x)^2}$$

$$\int_{2}^{5 \sinh x} \frac{5 \sinh x}{(3 + 2 \sinh x)^2}$$

$$\frac{5\sin x}{(3-2\cosh x)^2}$$

$$\frac{\sinh^2 x}{(2-3\sinh x)^2}$$

Question Number : 31 Question Id : 67809417054 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

If
$$y = \sqrt{\frac{1 - \cos x}{1 + \cos x}}$$
 then $\frac{dy}{dx}$ is

Options:

$$\sec^2\left(\frac{x}{2}\right)$$

$$\cos^2\left(\frac{x}{2}\right)$$

$$\frac{1}{2}\cos^2\left(\frac{x}{2}\right)$$

$$\frac{1}{2}\sec^2\left(\frac{x}{2}\right)$$

Question Number : 32 Question Id : 67809417055 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The angle between the curves $y = x^2 + 3x - 7$ and $y^2 = 2x + 5$ at (2,3) is

Options:

$$\tan \theta = 2$$

$$_2 \sec \theta = 2$$

$$\cos \theta = 1$$

$$\sin \theta = 3$$

Question Number: 33 Question Id: 67809417056 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The range of x for which the function $x^3 - 3x^2 - 45x + 2$ is increasing with x is

$$(3,-5)$$

$$_{2}$$
 $(-3,-5)$

Question Number: 34 Question Id: 67809417057 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The maximum value of the function $2x^3 - 12x^2 + 18x + 5$ is

Options:

- 1. 13
- 2. 12
- 3. 10
- 4 15

Question Number: 35 Question Id: 67809417058 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

If u is a homogeneous function of x and y with degree n then $x \frac{\partial u}{\partial x} + y \frac{\partial u}{\partial y} =$

Options:

- 1. nu
- $_2$ n^2u
- 3. nu
- $u^{2} + u^{2}$

Question Number : 36 Question Id : 67809417059 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The value of $\int \frac{\cos \sqrt{x}}{\sqrt{x}} dx$ is

$$2\sin\sqrt{x}+c$$

$$3\sin\sqrt{x}+c$$

$$2\sin x + c$$

$$\sin \sqrt{x} + c$$

Question Number : 37 Question Id : 67809417060 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The value of $\int \frac{dx}{\sqrt{a^2 - x^2}}$ is

Options:

$$\cos^{-1}\left(\frac{x}{a}\right) + c$$

$$\sin^{-1}\left(\frac{x}{a}\right) + c$$

$$\sinh^{-1}\left(\frac{x}{a}\right) + c$$

$$\sin^{-1}\left(\frac{a}{x}\right) + c$$

Question Number : 38 Question Id : 67809417061 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The value of $\int \frac{dx}{4x^2+4x+17}$ is

$$\frac{1}{8} \tan^{-1} \left(\frac{2x+1}{4} \right) + c$$

$$\frac{1}{4}\cot^{-1}\left(\frac{2x+1}{4}\right) + c$$

$$\frac{1}{8}\sin^{-1}\left(\frac{2x+1}{4}\right) + c$$

$$\int_{4}^{1} \tan^{-1}\left(\frac{2x+1}{4}\right) + c$$

Question Number : 39 Question Id : 67809417062 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The value of $\int \log x \, dx$ is

Options:

$$x \log x + x + c$$

$$2 x^2 \log x - x + c$$

$$x \log x - x + c$$

$$x\log x - \frac{x^2}{2} + c$$

Question Number: 40 Question Id: 67809417063 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\int_{1}^{4} \left(\sqrt{x} + \frac{1}{\sqrt{x}} \right) dx$ is

Options:

$$-\frac{20}{3}$$

Question Number: 41 Question Id: 67809417064 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The value of $\int_0^{\pi/2} \sin^2 x \, dx$ is

$$\frac{\pi}{2}$$

$$-\frac{\pi}{4}$$

$$\frac{\pi}{4}$$

Question Number: 42 Question Id: 67809417065 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The area enclosed between the curve $y^2 = 4ax$ and the line x = 2y is

Options:

$$\frac{64}{5}$$
 sq. units

$$\frac{64}{3}$$
 sq. units

$$\frac{65}{4}$$
 sq. units

$$\frac{63}{4}$$
 sq. units

Question Number : 43 Question Id : 67809417066 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The value of $\lim_{n\to\infty} \left[\frac{1}{n+1} + \frac{1}{n+2} + \cdots + \frac{1}{n+n} \right]$ is

Options:

$$_4 \log n$$

Question Number: 44 Question Id: 67809417067 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Form the differential equation by eliminating the arbitrary constant a from $ay^2 = x^3$ Options:

$$\frac{dy}{dx} = \frac{3y}{2x}$$

$$\frac{dy}{dx} = \frac{2x}{3y}$$

$$\frac{dy}{dx} = \frac{x}{y}$$

$$\frac{dy}{dx} = \frac{2y}{x}$$

Question Number: 45 Question Id: 67809417068 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The solution of $\sqrt{1-y^2}dx + \sqrt{1-x^2}dy = 0$ is

Options:

$$\cos^{-1} x + \cos^{-1} y = c$$

$$\int_{\gamma} \sinh^{-1} x + \cosh^{-1} y = c$$

$$\cos^{-1} x + \sec^{-1} x = c$$

$$\sin^{-1} x + \sin^{-1} y = c$$

Question Number : 46 Question Id : 67809417069 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The solution of $\frac{dy}{dx} = (4x + y + 1)^2$ is

$$\int_{1}^{\infty} \frac{1}{2} \tan^{-1} \left(\frac{4x + y + 1}{2} \right) = x + c$$

$$\int_{2}^{1} \cot^{-1}\left(\frac{4x+y+1}{2}\right) = x + c$$

$$\int_{3}^{2} -\frac{1}{2} \tan^{-1} \left(\frac{4x+y+1}{2} \right) = x + c$$

$$\frac{1}{2}\tan^{-1}\left(\frac{4x-y-1}{2}\right) = x + c$$

Question Number: 47 Question Id: 67809417070 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The solution of exact differential equation $2xy dx + x^2 dy = 0$ is

Options:

$$_{1.} x^{2}y^{2} = c$$

$$_2$$
 $x^2y = c$

$$_{3.}x^{3}y=c$$

$$_{4.} x^2 y^3 = c$$

Question Number: 48 Question Id: 67809417071 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The solution of $\frac{dy}{dx} + y = e^{-x}$ is

Options:

$$\int_{1}^{\infty} (x+c)e^{-x}$$

$$(x-c)e^x$$

$$(x+c)e^x$$

$$(x+c)e^{-2x}$$

Question Number : 49 Question Id : 67809417072 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

The particular integral of $(D^2 + 5D + 6)y = e^x$ is

$$\frac{-e^{-x}}{12}$$

$$\frac{e^{2x}}{12}$$

$$\frac{e^{\lambda}}{12}$$

Question Number: 50 Question Id: 67809417073 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The complementary function of $(D^2 + 3D + 2)y = 8sin5x$ is

Options:

$$c_1e^{-x}+c_2e^{-2x}$$

$$c_1 e^x + c_2 e^{2x}$$

$$_{3.}$$
 $c_1e^{-x}+c_2e^{2x}$

$$c_1e^{2x}+c_2e^{3x}$$

Physics

Number of Questions:25Display Number Panel:YesGroup All Questions:No

Question Number: 51 Question Id: 67809417074 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Which of the following is not the unit of energy?

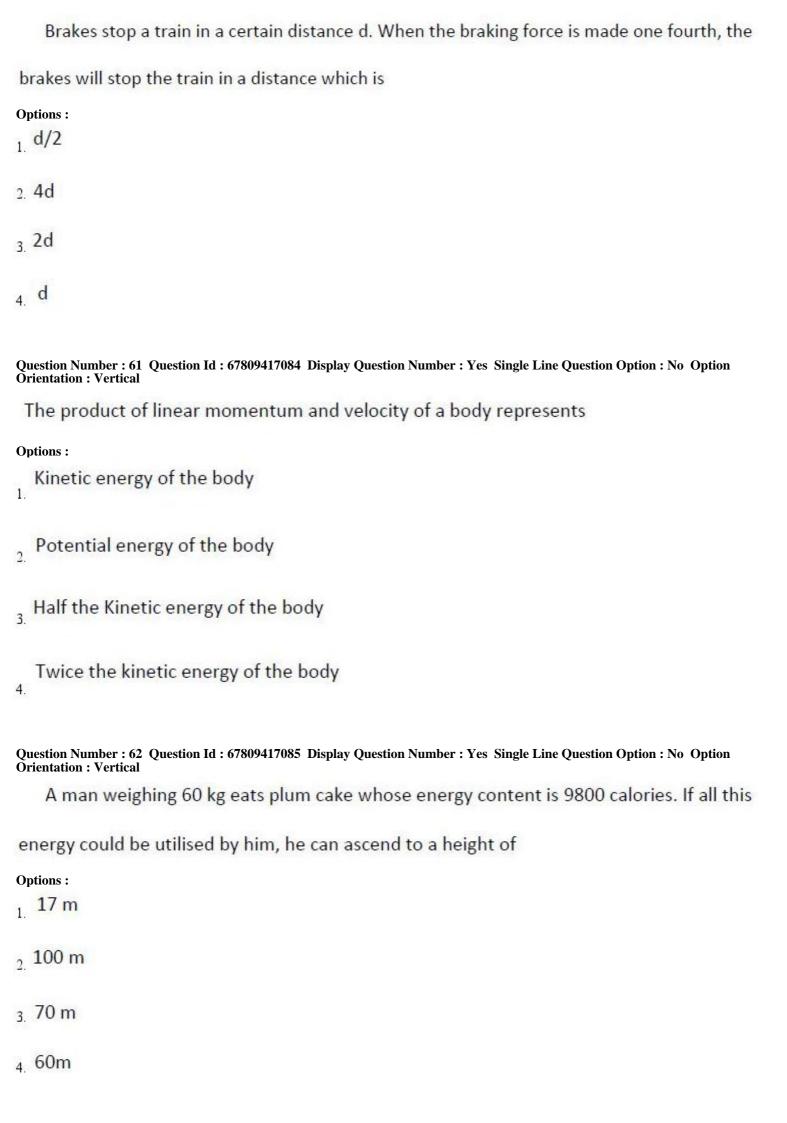
- watt second
- 2. Pascal metre

Newton metre 4 Kilowatt hour Question Number: 52 Question Id: 67809417075 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** The height of Mercury barometer is 76 cm and density of Mercury is 13.6 g/cc. The corresponding height of water barometer in SI system is **Options:** 10.336 m ₂ 103.36 m 3.6m 4 1.0336 m Question Number: 53 Question Id: 67809417076 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** Angle made by the vector $(\sqrt{3} \ \overline{i} + \overline{j})$ with the X-axis is **Options:** $1 \pi/2$ $_{2}$ $\pi/4$ $_{4}$ $\pi/6$ Question Number: 54 Question Id: 67809417077 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** The minimum number of unequal forces in a plane that can keep a particle in equilibrium is **Options:**

2. 2
3. 3
4. 6
Question Number: 55 Question Id: 67809417078 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
A body is thrown with a velocity of $(4\bar{i}+3\bar{j})$ m/s. The maximum height attained by
the body is (g=10 ms ⁻²)
Options:
2.5 m
2. 4.5 m
3. 0.8 m
4. 0.45 m
Question Number: 56 Question Id: 67809417079 Display Question Number: Yes Single Line Question Option: No Option
Orientation : Vertical
A person in a lift, which ascends up with acceleration 10ms ⁻² , drops a stone from a height of
10m. The time of descent is (g=10 ms ⁻²)
Options:
1. 0.5 s
2. 1 s
3. 1.5 s
4. ^{2 s}
Question Number: 57 Question Id: 67809417080 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
For a projectile, the ratio of maximum height reached to the square of time of flight is

1. 5:4
_{2.} 5:2
3. 5:1
4. 10:1
Question Number : 58 Question Id : 67809417081 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
The ratio of distances travelled by a body, starting from rest and travelling with uniform
acceleration, in successive intervals of time of equal duration will be
Options:
1. 1:2:3
_{2.} 1 :4:9
_{3.} 1 :3:5
4. 1:9:16
Question Number: 59 Question Id: 67809417082 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
A force of 12 N acts on a body of mass 4 kg placed on a rough surface. The coefficient of
friction between body and surface is 0.2 and take g= 10 ms ⁻² . The acceleration of the body in
ms ⁻² is
Options:
1. 1
2. 0.5
3. 0.25
4. Zero
Question Number • 60 Question Id • 67809417083 Display Question Number • Ves Single Line Question Ontion • No Ontion

Question Number: 60 Question Id: 67809417083 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical



Question Number : 63 Question Id : 67809417086 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

A crane can lift up 10,000 kg of coal in 1 hour from a mine of depth 180m. If the efficiency of

the crane is 80%, its input power must be (g=10 ms⁻²)

Options:

- 62.5 kW
- ₂ 6.25 kW
- 3. 50 kW
- 4. 5 kW

Question Number: 64 Question Id: 67809417087 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The graph of acceleration as a function of displacement in the case of a body executing simple harmonic motion is

Options:

- Parabola
- ₂ Hyperbola
- Straight line with positive slope
- Straight line with negative slope

Question Number: 65 Question Id: 67809417088 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The pendulum of length 'L' swings from mean position to mean position 'n' times in one second. The value of acceleration due to gravity is

- $1. \pi^2 n^2 L$
- $_{2}$ $2\pi^{2}n^{2}L$

- $_{3.}(\pi^2n^2L)/2$
- $_4$ $4\pi^2 n^2 L$

Question Number: 66 Question Id: 67809417089 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

When a source of sound is in motion towards a stationary observer, the effect observed is Options:

- Decrease in velocity of sound
- , Increase in velocity of sound
- increase in frequency of sound
- 4 decrease in frequency of sound

Question Number: 67 Question Id: 67809417090 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The voice of a male person is different from that of a female person because

Options:

- , Two sounds have different phases
- Two persons are of different size
- Two sounds travel with different velocities
- 4 Two sounds have different pitch

Question Number : 68 Question Id : 67809417091 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

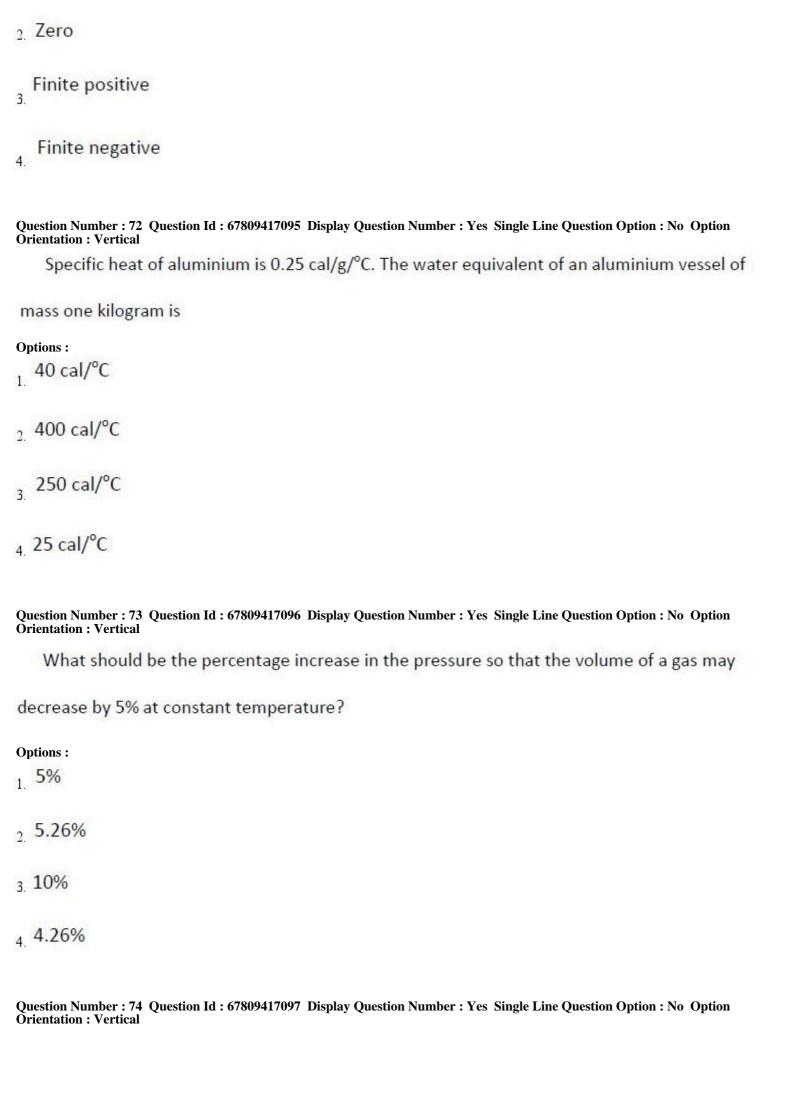
If the sound absorption of a hall is changed by 2%, then the percentage change in the

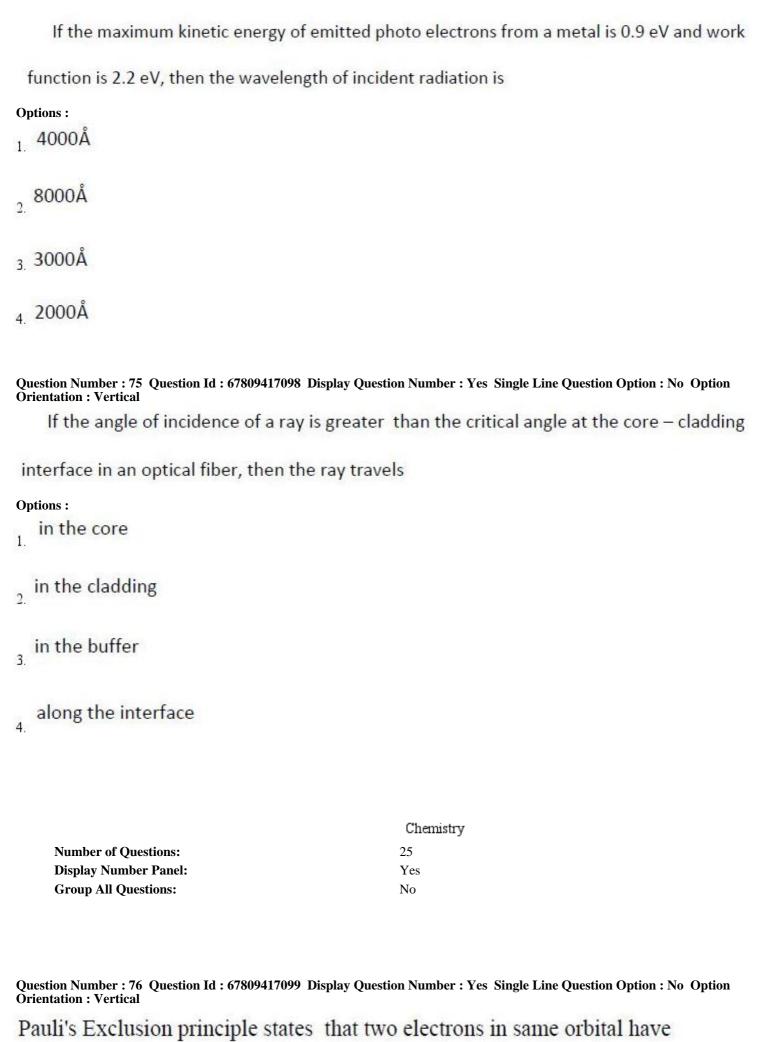
reverberation time is

Options:

1. 2%

2. 4%
3. 1 %
No change
Question Number : 69 Question Id : 67809417092 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
In which of the following process, the internal energy of the system remains constant?
Options:
1. Adiabatic
2. Isothermal
3. Isobaric
4. Isochoric
Question Number : 70 Question Id : 67809417093 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
Heat required to raise the temperature of one gram of water through 1 K is
Options:
1. 1.0 Kcal
2. 0.1 Kcal
3. 0.01 Kcal
4. 0.001 Kcal
Question Number : 71 Question Id : 67809417094 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
The specific heat of a gas in an isothermal process is
Options: 1. infinity





same spins

different spins

opposite spins

vertical spins

Question Number: 77 Question Id: 67809417100 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Orbits in which electrons move according to Bohr are

Options:

1. elliptical

2 cylindrical

3. circular

4 oval

Question Number: 78 Question Id: 67809417101 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Phosphorus has an atomic number of 15. A stable phosphorus atom has an electronic configuration of

Options:

$$1s^22s^22p^63p^5$$

$$_{2}$$
 1s²2s²2p⁶3s²3p³

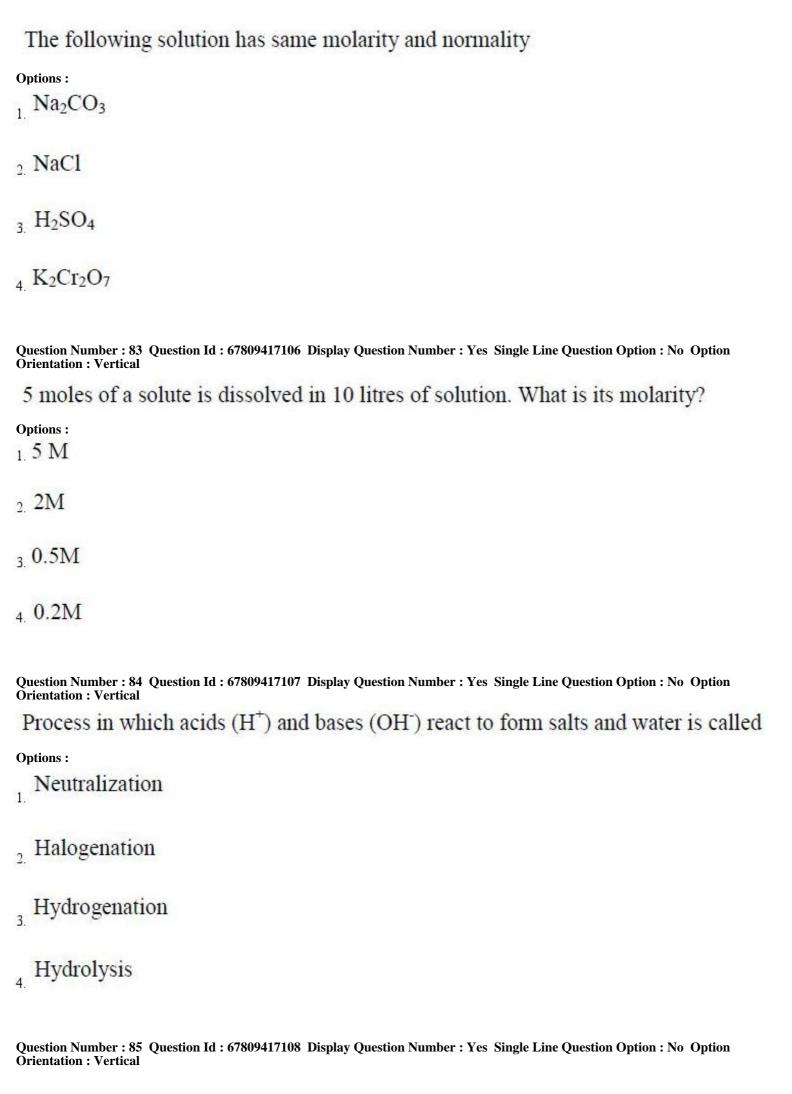
$$_{3}$$
 $1s^22s^22p^63s^23p^14s^2$

$$1s^21p^61d^7$$

Question Number: 79 Question Id: 67809417102 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

NaCl is classified as having what kind of bonds in the solid phase?
Options:
1. Covalent
2. Ionic
3. Polar
4. vander Waals
Question Number: 80 Question Id: 67809417103 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The Bond formed due to sharing of electrons is
Options:
1. Ionic bond
2. Metallic bond
3. Polar bond
4. Covalent bond
Question Number: 81 Question Id: 67809417104 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The normality of solution obtained by dissolving $5.3~{\rm grams}$ of ${\rm Na_2CO_3}$ in $1~{\rm litresolution}$ is
Options:
1. 1N
2. 0.1N
3. 0.05N
4. 0.5N

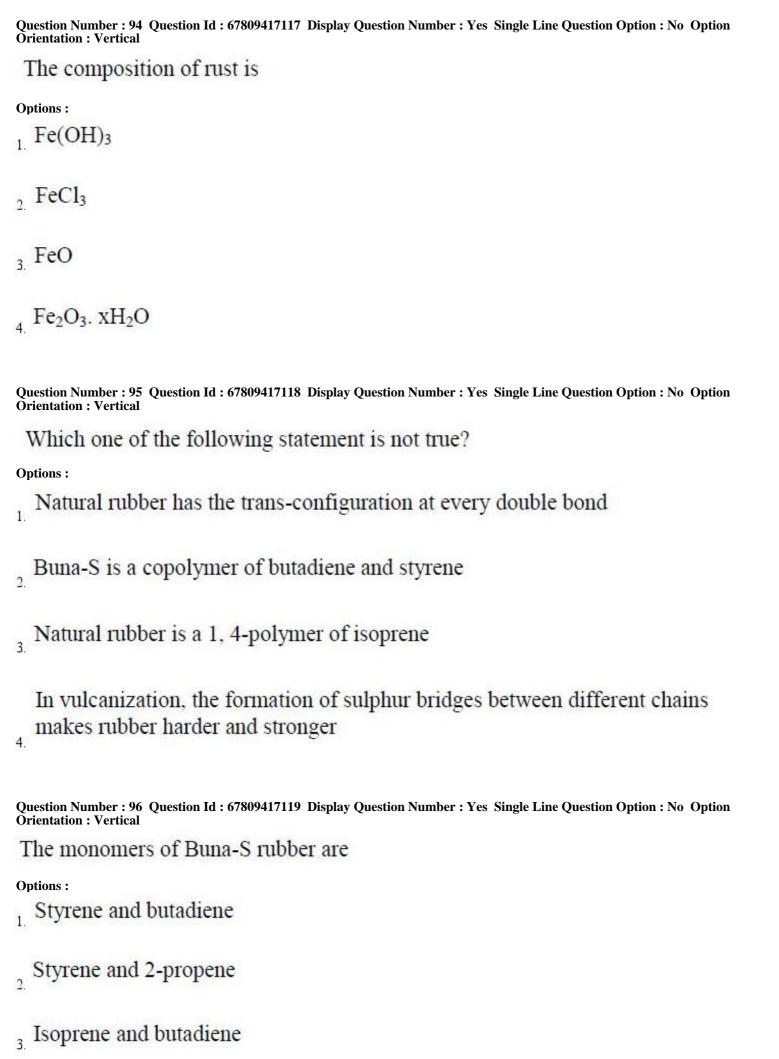
 $Question\ Number: 82\ Question\ Id: 67809417105\ Display\ Question\ Number: Yes\ Single\ Line\ Question\ Option: No\ Option\ Orientation: Vertical$



A substance that donates a pair of electrons to form coordinate covalent bond is called
Options:
1. Lewis acid
2. Lewis base
3. Bronsted-Lowry acid
Bronsted-Lowry base
Question Number: 86 Question Id: 67809417109 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
One Faraday is equal to
Options: 1. 99650 C
_{2.} 93100 C
_{3.} 96500 C
_{4.} 94500 C
Question Number: 87 Question Id: 67809417110 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The cell reaction of a cell is $Mg(s) + 2 H^{+}(aq) \rightarrow Mg^{2+}(aq) + H_{2}(g)$. If the standard reduction potential of Zn is -2.372 V , then the emf of the cell is
Options:
_{1.} +2.372 V
_{2.} – 2.372 V
3. 0.00 V
_{4.} -1.372 V
Question Number: 88 Question Id: 67809417111 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Galvanic cells are the cells which convert
Options:
Electrical energy to chemical energy
2. Chemical energy to electrical energy
Chemical energy to free energy
4. Potential energy to kinetic energy
Question Number: 89 Question Id: 67809417112 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Mass of substance produced at electrode is directly proportional to the quantity of electricity passed. This is known as
Options:
Faraday's second law
Faraday's first law
Newton's third law
Newton's first law
Question Number : 90 Question Id : 67809417113 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
Hardness of water is expressed in terms of equivalent of
Options: Na_2CO_3
$_{2}$ $K_{2}CO_{3}$
$_{3.}$ MgCO ₃
4. CaCO ₃

Question Number : 91 Question Id : 67809417114 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
Temporary hardness is caused by
Options:
Carbonates of calcium and magnesium
Chlorides of calcium and magnesium
Sulphates of calcium and magnesium
4. Nitrates of Calcium
Question Number : 92 Question Id : 67809417115 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
The exhausted zeolite bed can be regenerated by washing with
Options:
1. NaCl
_{2.} dil. NaOH
3 dil. HCl
3. (CONTENT OF THE TOTAL OF THE
4. Distilled water
Question Number: 93 Question Id: 67809417116 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Corrosion is an example of
Options:
1. Oxidation
2. Reduction
Electrolysis 3.
4. Halogenation



Styrene and sulphur

Question Number: 97 Question Id: 67809417120 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The plastics which soften when heat is applied with or without pressure, but require cooling to set them to shape are called as

Options:

- Thermosofting materials
- Thermosetting materials
- Thermoplastic materials
- Thermostatting materials

Question Number: 98 Question Id: 67809417121 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Which one of the following statement is not true about ideal fuel?

Options:

- High calorific value
- , High moisture content
- 3 Low cost
- Moderate ignition temperature

Question Number: 99 Question Id: 67809417122 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Environmental pollution affects

- Humans only
- 2 Plants only

Biotic components
Both abiotic and biotic components
Question Number: 100 Question Id: 67809417123 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Layer of atmosphere in which ozone layer lies is
Options: Troposphere 1.
2. Stratosphere
Exosphere 3.
4. Mesosphere
Chemical Engineering Number of Questions: 100 Display Number Panel: Group All Questions: No
Question Number: 101 Question Id: 67809417124 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Ethane gas (molecular weight = 30 amu) is stored in a cylinder at a temperature of 27°C and a pressure of 1 bar. Take universal gas constant as 8 kJ/(kmol.K). Considering ethane to be ideal gas, the density is kg/m ³ .
Options:
1. 5/36
2. 1/80
3. 5/4
4 1250

Question Number: 102 Question Id: 67809417125 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** Which communition equipment uses critical speed concept? 1. Hammer mill Attrition mill 3. Jaw crusher Ball mill Question Number: 103 Question Id: 67809417126 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** In all binary phase diagrams a two-phase region separates single phase regions. This is known as **Options:** Lever rule 2. Kirkendall effect 1-2-1 rule Ficks law Question Number: 104 Question Id: 67809417127 Display Question Number: Yes Single Line Question Option: No Option Which of the following is used to measure Point velocity of a fluid?

Options:

- 1. Rotameter
- Pitot Tube
- Venturimeter
- Coriolis meter

Question Number: 105 Question Id: 67809417128 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical**

The resistance-temperature relation for platinum elements is given by

Options:

- Cox chart
- Callendar equation
- 3. Trouton's rule
- Planck's law

Question Number: 106 Question Id: 67809417129 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical**

Which industry releases nitrogen oxides into atmosphere?

Options:

- Nuclear power
- 2. Agriculture
- Stone crushing
- Petroleum refining

Question Number: 107 Question Id: 67809417130 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical**

A gas mixture contains only methane (molecular weight = 16 amu) and ethane (molecular weight = 30 amu). The composition of ethane is 20 percent by volume. Assuming the application of ideal gas law, the average molecular weight of the gas mixture is

Options: 1.23 2.18.8 3.14.6 4.46 Question Number: 108 Question Id: 67809417131 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** Binary gas diffusion coefficients for gas mixtures are in between m^2/s . **Options:** 1. 10⁻² and 10⁻⁴ 2. 10⁻⁴ and 10⁻⁶ 3. 10⁻⁶ and 10⁻⁸ 4. 10⁻⁸ and 10⁻¹⁰ Question Number: 109 Question Id: 67809417132 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** The degree of polymerization in polyethylene is 2000. Molecular weight of ethylene is 28 amu. The average molecular weight of polyethylene is **Options:** 1.2000 2. 28000 3.56000 4. 28 Question Number: 110 Question Id: 67809417133 Display Question Number: Yes Single Line Question Option: No Option **Orientation: Vertical** Duralumin is an alloy of **Options:** 1. Aluminium and copper Aluminium and deuterium 3. Aluminium and duron Aluminium and chromium Question Number: 111 Question Id: 67809417134 Display Question Number: Yes Single Line Question Option: No Option **Orientation**: Vertical Sacrificial anode is used for **Options:** Anodic protection 2. Cathodic protection 3. Passivation

Question Number: 112 Question Id: 67809417135 Display Question Number: Yes Single Line Question Option: No Option
Orientation : Vertical Colour measurement can be done by
Options:
1. Optical pyrometer
2. Coriolis meter
3. Spectrometer
4. X-ray diffraction
T. Ze lay unification
Question Number: 113 Question Id: 67809417136 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Enthalpy change during a throttling process is
Options:
1. Negative
2. Positive
3. Zero
4. Cannot be determined
Question Number: 114 Question Id: 67809417137 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
A Carnot engine operates with a heat source reservoir at 127°C and a heat sink reservoir at 27°C. What is its thermal efficiency?
Options:
1. 0.25
2. 0.75
3. 0.50
4. 0.20
Question Number: 115 Question Id: 67809417138 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Actual pressure of isobutylene is 10 bar and its critical pressure is 40 bar. Then reduced pressure is
Options:
1. 30 bar
2. 4 bar
3. 0.25 bar
4. 50 bar
Question Number: 116 Question Id: 67809417139 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Activated studge process is employed in the treatment of waste waters in

4. Magnetization

1. Primary treatment
2. Secondary treatment
3. Tertiary treatment
4. Ion-exchange
Question Number: 117 Question Id: 67809417140 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
An example for path function is
Options:
1. Pressure
2. Volume
3. Work
4. Temperature
Question Number: 118 Question Id: 67809417141 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Absorption is the mass transfer operation that occurs between phases.
Options:
1. Gas and gas
2. Gas and liquid
3. Liquid and solid
4. Liquid and liquid
Question Number: 119 Question Id: 67809417142 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Which of the following industries uses blast furnace?
Options:
1. Hindustan Petroleum Corporation Limited
2. Visakhapatnam Steel Plant
3. Coramandel Fertilizers Limited
4. Godavari Fertilizers Limited
Question Number: 120 Question Id: 67809417143 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The degree of freedom when ice, water and water vapour coexist in equilibrium is
Options:
11
2. Zero
3. 1
4. 3
Question Number: 121 Question Id: 67809417144 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
A refrigeration system requires 2 kilowatt power for a refrigeration rate of 5 kilojoules per second. What is its coefficient of performance?

Options:
1. 0.4

2. 0.6
3. 2.5
4. 3
Question Number: 122 Question Id: 67809417145 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
In which reactor the residence time is same for all elements of fluid?
Options:
1. Batch reactor
2. Ideal plug flow reactor
3. CSTR
4. Packed bed reactor
Question Number: 123 Question Id: 67809417146 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The fluids for which the ratio of shear stress to the shear rate is constant are called as fluids.
Options:
1. Dilatant
2. Bingham
3. Newtonian
4. Thixotropic
Question Number: 124 Question Id: 67809417147 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Film theory in interphase mass transfer is described by
Options:
1. Lewis and Whitman
2. Treybal
3. Dankwerts
4. Henry
Question Number: 125 Question Id: 67809417148 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
In a chemical reaction the thermodynamic equilibrium constant is unaffected by
Options:
1. Temperature
2. Concentration of materials
3. Conversion of reactants
4. Presence of inerts
Question Number: 126 Question Id: 67809417149 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The temperature of a furnace is raised at the rate of 5°C per minute. The thermometer that is used to measure the furnace
temperature is a first order system with a time constant of 2 min. The maximum dynamic error would be°C.
Options:

2. 5
3. 10
4. 25
Question Number: 127 Question Id: 67809417150 Display Question Number: Yes Single Line Question Option: No Optio Orientation: Vertical
Antoine equation is used to compute at a given temperature.
Options:
1. Atomic radius
2. Vapour pressure
3. Bond energy
4. Wave length
Question Number: 128 Question Id: 67809417151 Display Question Number: Yes Single Line Question Option: No Optio Orientation: Vertical
A gas is confined in a vessel with rigid and insulating walls. This is an example of system.
Options:
1. Closed
2. Open
3. Isolated
4. Extensive
Question Number: 129 Question Id: 67809417152 Display Question Number: Yes Single Line Question Option: No Optio Orientation: Vertical
Entropy of the universe is
Options:
1. Constant
2. Decreasing
3. Increasing
4. Negative
Question Number: 130 Question Id: 67809417153 Display Question Number: Yes Single Line Question Option: No Optio Orientation: Vertical
The temperature at which the equilibrium vapour pressure of the liquid is equal to the existing partial pressure of the vapour termed as of the mixture.
Options:
1. Normal boiling point
2. Critical temperature
3. Bubble point
4. Dew point
Question Number: 131 Question Id: 67809417154 Display Question Number: Yes Single Line Question Option: No Optio Orientation: Vertical
An ideal gas undergoes a polytropic change which can be represented by PV^{δ} =
constant. If δ is taken as infinity then the path that is followed by the gas is

1. Isothermal
2. Isobaric
3. Adiabatic
4. Isovolumic
Question Number: 132 Question Id: 67809417155 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
An example for extensive property in thermodynamics is
Options:
1. Pressure
2. Temperature
3. Enthalpy
4. Density
Question Number: 133 Question Id: 67809417156 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
In sedimentation operation, which of the following settling takes place at final stages?
Options:
1. Free Settling
2. Compressive Settling
3. Hindered Settling
4. Brownian Settling
Question Number: 134 Question Id: 67809417157 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The ratio of area of opening of two successive Tyler screens is
Options:
$1. 2^{1/2}$
2. 1
3. 2.
4. 4
Question Number: 135 Question Id: 67809417158 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The tower internals that are difficult to clean are

1. Perforated trays
2. Valve trays
3. Random packing
4. Structured packing
Question Number : 136 Question Id : 67809417159 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
states that in conduction heat flux is proportional to the temperature gradient.
Options:
l. Planck's law
2. Stefan-Boltzmann law
3. Fourier law
4. Kirchhoff law
Question Number: 137 Question Id: 67809417160 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
In a shell and tube heat exchanger, to promote cross flow and raise the average velocity of the shell side fluid, are installed in the shell side.
Options : 1. Nozzles
2. Tie rods
3. Tube sheets
4. Baffles
T. Dellies
Question Number: 138 Question Id: 67809417161 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Cross-ambient effect is associated with thermometer.
Options:
1. Bimetallic
2. Vapour-actuated
3. Resistance
4. Mercury-in-glass
Question Number: 139 Question Id: 67809417162 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
An example for non-Newtonian fluid is
Options:
1. Water
2. Alcohol
3. Benzene
4. Toothpaste
Question Number: 140 Question Id: 67809417163 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Continuous vacuum filter ordinarily uses mm Hg vacuum.
Options:
1. Less than 250
2. 250
3 250 - 500

4. Above 500
Question Number: 141 Question Id: 67809417164 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Units of thermal conductivity are
Options:
$_{1.}$ W/(m.K)
_{2.} (W.m)/K
$W/(m^2.K)$
$_{4.}$ (W.m ²)/K
Question Number: 142 Question Id: 67809417165 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Filter aid is used to
Options: 1. Increase the rate of filtration
2. Decrease the pressure drop
Increase the porosity of the cake
4. Support base for the system
Support dans 15. the system
Question Number: 143 Question Id: 67809417166 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Decanters are used in the operation.
Options:
1. Distillation
2. Extraction
3. Absorption
4. Drying
Question Number: 144 Question Id: 67809417167 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
For monatomic gases, the ratio of heat capacities $C_p/C_V =$
Options:
1. 1.0
2. 1.25

3. 1.4

4. 1.67
Question Number: 145 Question Id: 67809417168 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The Power Law which makes use of work index is
Options:
1. Kick's law
2. Bond's law
3. Rittinger's law
4. Fick's law
Question Number: 146 Question Id: 67809417169 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The size reduction mechanism in Jaw crushers is
Options:
1. Attrition
2. Compression
3. Cutting
4. Impact
Question Number: 147 Question Id: 67809417170 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Air borne suspension of solid or liquid is known as
Options:
1. Aerosol
2. Cresol
3. Xylol
4. Dirt
Question Number: 148 Question Id: 67809417171 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Absolute pressure at a point is 5.0 bar and surrounding pressure is 2.0 bar. The
gauge pressure is bar.
Options:
1. 7.0
2. 3.0
3. 2.5

4. 0.4

Raoult's law is applicable for the case of operation.
Options:
1. Extraction
2. Distillation
3. Absorption
4. Drying
Question Number: 150 Question Id: 67809417173 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
type of tower is used for gas dispersed in the liquid in the form of bubbles.
Options: 1. Packed tower
2. Spray tower
3. Venture scrubber
4. Tray tower
Question Number: 151 Question Id: 67809417174 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Which type of heat exchangers are generally used in diary and food industries?
Options:
1. Shell and tube
2. Double pipe
3. Plate type
4. Extended surface
Question Number: 152 Question Id: 67809417175 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Reynolds number is defined as
Options:
1. (inertial force)/(gravity force)
2. (gravity force)/(inertial force)
3. (inertial force)/(viscous force)
4. (viscous force)/(inertial force)
Question Number: 153 Question Id: 67809417176 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
If the viscosity is 1 centipoise, diffusivity is 1 cm ² /s and density is 1 g/cm ³ , then
the Schmidt number is
Options:
1. 1
2. 0.1

4. 0.001
Question Number: 154 Question Id: 67809417177 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical $ Let \ h = heat \ transfer \ coefficient, \ D = diameter, \ k = thermal \ conductivity, $
then Nusselt number is given by
Options:
(hD)/k
_{2.} (hk)/D
$_{3.}$ (kD)/h
4. D/(kh)
Question Number: 155 Question Id: 67809417178 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical For concentration of highly heat sensitive materials such as orange juice evaporator is used. Options: 1. Long-tube climbing film 2. Falling film type 3. Forced circulation 4. Agitated film
Question Number : 156 Question Id : 67809417179 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
An instrument is composed of four parts. Each part has a specified accuracy of
±0.25 percent. The least accuracy is percent.
Options:
1. ±1.0
2. ±2.0
_{3.} ±0.25
4. ±0.5

 $Question\ Number: Yes\ Single\ Line\ Question\ Option: No\ Option\ Orientation: Vertical$

Main constituents of biogas are
Options:
Methane and carbon dioxide
2. Propane, butane and nitrogen oxide
3. Butane, pentane and sulphur dioxide
4. Carbon dioxide and water
Question Number: 158 Question Id: 67809417181 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The temperature at which ignition will occur without the presence of a spark or flame is known as
Options:
1. Adiabatic reaction temperature
2. Autoignition temperature
3. Theoretical flame temperature
4. Fire point
Question Number: 159 Question Id: 67809417182 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Fault-tree analysis is a
Options:
1. Hazard assessment technique
2. Biochemical analysis method
3. Tree trunk morphology study
4. Combustion based activity
Question Number: 160 Question Id: 67809417183 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Bhopal disaster was due to
Options:
1. Methyl isocynate
2. Methyl cynate
3. Isomethyl cynate
4. Potassium cyanide
Question Number: 161 Question Id: 67809417184 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Which property of the fluid is a measure of its resistance to flow?
Options:
1. Density
2. Viscosity
3. Vapour pressure
4. Surface tension
Question Number: 162 Question Id: 67809417185 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The equation that is applicable to the flow of homogeneous fluid through packed beds is
Options:
1. Darcy
2. Bernoutli

4. Ergun
Question Number: 163 Question Id: 67809417186 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
In evaporator design, steam economy is defined as
Options:
1. Mass of water evaporated per unit time
2. Mass of steam fed per unit time
3. Mass of water evaporated per mass of steam fed
4. Mass of steam fed per mass of water evaporated
Question Number: 164 Question Id: 67809417187 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Recovery of penicillin from fermentation broth is the application of type of mass transfer operation.
Options:
1. Absorption
2. Distillation
3. Extraction
4. Adsorption
Question Number : 165 Question Id : 67809417188 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical
Double pipe heat exchangers are used when the heat transfer areas are ft².
Options:
1. More than 1000
2. Between 100 and 150
3. Between 200 and 500
4. Between 500 and 1000
Question Number: 166 Question Id: 67809417189 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Bottoms from the beer still are known as
Options:
1. Molasses
2. Slops
3. Bagasse
4. Ethanol
Question Number: 167 Question Id: 67809417190 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Burning of solid wastes at high temperatures is known as
Options:
1. Flocculation
2. Incineration
3. Digestion
4. Calcination

3. Hagen-poisueille

Question Number: 168 Question Id: 67809417191 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The final products of combustion of hydrocarbons are

Options:

- 1. Hydrogen and carbon
- 2. Carbon dioxide and water
- Carbon, carbon monoxide, carbon dioxide, hydrogen
- Carbon, carbon dioxide, hydrogen

Question Number : 169 Question Id : 67809417192 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Which of the following resembles natural rubber?

Options:

- Polythene
- Polyisoprene
- 3. PVC
- Polystyrene

Question Number: 170 Question Id: 67809417193 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

India's coal production is mainly in

Options:

- 1. Jharkhand
- 2. Andhra Pradesh
- Kerala
- 4. Goa

Question Number: 171 Question Id: 67809417194 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

An example for solid fuel is

Options:

- 1. Naphtha
- 2. LPG
- 3. Coal
- 4. Granite

Question Number : 172 Question Id : 67809417195 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Coal gasification can be done using

Options:

- Lurgi process
- United German process
- Unilever process
- 4. Coal Reich process

Question Number : 173 Question Id : 67809417196 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Energy audits help in improving

1. Energy consumption
2. Energy conversion
3. Energy conservation
4. Energy dissipation
Question Number: 174 Question Id: 67809417197 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Which of the following is a residue product from petroleum refinery?
Options:
1. LPG
2. Petrol
3. Diesel
4. Grease
Question Number: 175 Question Id: 67809417198 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The amount of moisture in a solid that exerts a vapour pressure less than the normal vapour pressure of water at the given temperature is called as
Options:
1. Bound moisture
2. Unbound moisture
3. Free moisture
4. Equilibrium moisture
Question Number: 176 Question Id: 67809417199 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
An example for variable area meter is
Options:
1. Rotameter
2. Orifice meter
3. Venturi meter
4. Wet gas meter
Question Number: 177 Question Id: 67809417200 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Which of the following is not a function of an instrument?
Options:
1. Registering
2. Indicating
3. Drifting
4. Recording
Question Number: 178 Question Id: 67809417201 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Channeling is an undesirable operation in
Options:
1. Packed towers
2. Spray towers
3. Venture scrubbers

4. Tray towers
Question Number: 179 Question Id: 67809417202 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Unit for overall heat transfer coefficient is
Options:
$W/(m^2.K)$
₂ W/(m.K)
3. (W.m)/K
$_{4.}$ (W.m ²)/K
Question Number: 180 Question Id: 67809417203 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
An example for toxic metal is
Options:
1. Calcium
2. Sodium
3. Potassium
4. Mercury
Question Number: 181 Question Id: 67809417204 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Cyclone separator is used for removing from a gas.
Options:
1. Carbon monoxide
2. Sulphur dioxide
3. Nitrogen oxides
4. Particulate matter
Question Number: 182 Question Id: 67809417205 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
The calibrated range of a thermometer is 200°C to 500°C. The span of the
thermometer is
Options:
_{1.} 200°C
_{2.} 500°C

4. 250°C
Question Number: 183 Question Id: 67809417206 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical operation is carried out for economical removal of phenol from aqueous wastes.
Options:
1. Distillation
2. Extraction
3. Absorption
4. Drying
Question Number: 184 Question Id: 67809417207 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Handling of corrosive liquids can be done using pump.
Options:
1. Piston
2. Plunger
3. Gear
4. Diaphragm
Question Number: 185 Question Id: 67809417208 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical oil is an edible oil.
Options:
1. Lube
2. Linseed
3. Rice bran
4. Castorseed
T. Gasturseed
Question Number: 186 Question Id: 67809417209 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Saponification process is used in the manufacture of
Options:
1. Soaps
2. Soda ash
3. Soda lime glass
4. Varnishes
Question Number: 187 Question Id: 67809417210 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Bypassing a fluid stream by splitting it into two parallel streams is done for the purpose of achieving
Options:
Accurate control in concentration
2. Maximum yield
3. Optimum conversion

3. 300°C

Isothermal conditions

Question Number : 188 Question Id : 67809417211 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Contact process is used for manufacture of

Options:

- 1. Nitric acid
- 2. Hydrochloric acid
- 3. Phosphoric acid
- 4. Sulphuric acid

Question Number : 189 Question Id : 67809417212 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Synthesis gas is a mixture of

Options:

- 1. Carbon monoxide and hydrogen
- 2. Water vapour and carbon monoxide
- 3. Water vapour and carbon dioxide
- Carbon dioxide and hydrogen

Question Number: 190 Question Id: 67809417213 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Sylvinite is a mineral of

Options:

- 1. Iron
- 2. Potassium
- 3. Copper
- 4. Silver

Question Number: 191 Question Id: 67809417214 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

The catalyst used in the synthesis of ammonia from nitrogen and hydrogen is

Options:

- 1. Cuprous oxide
- 2. Iron oxide
- 3. Platinum
- 4. Nickel

Question Number: 192 Question Id: 67809417215 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

Prilling process is required in the manufacture of

Options:

- 1. Industrial gases
- 2. Urea
- 3. Glucose
- Nitric acid

Question Number: 193 Question Id: 67809417216 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical

IMI process for phosphoric acid production employs
Options:
1. Hydrochloric acid leaching
2. Nitric acid leaching
3. Sulphuric acid leaching
4. Electric arc furnace
Question Number: 194 Question Id: 67809417217 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Solvay process is used in the production of
Options:
1. Soda ash
2. Coal tar
3. Lime soda
4. Gypsum
Question Number: 195 Question Id: 67809417218 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Hardness to water is caused by
Options:
1. Dissolved salts of calcium and magnesium
2. Dissolved gases
3. Insoluble matter
4. Pathogenic organisms
Question Number: 196 Question Id: 67809417219 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Orientation: Vertical Green pigment is Options:
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters.
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options:
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles 3. Dissolved salts
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles 3. Dissolved salts
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles 3. Dissolved salts 4. Dissolved gases Question Number: 198 Question Id: 67809417221 Display Question Number: Yes Single Line Question Option: No Option
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles 3. Dissolved salts 4. Dissolved gases Question Number: 198 Question Id: 67809417221 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical The forces acting on a settling particle are Options:
Orientation: Vertical Green pigment is Options: 1. Titanium dioxide 2. Chromium compound 3. Lithopone 4. Iron oxide Question Number: 197 Question Id: 67809417220 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical Coagulation is used for the removal of from waste waters. Options: 1. Bacteria 2. Suspended particles 3. Dissolved salts 4. Dissolved gases Question Number: 198 Question Id: 67809417221 Display Question Number: Yes Single Line Question Option: No Option Orientation: Vertical The forces acting on a settling particle are

- 3. Centrifugal, electrostatic, shear
- 4. Compressive, shear, tensile

Question Number : 199 Question Id : 67809417222 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

In an experimental study, combustion of 1 mol of methane gas is carried out under adiabatic conditions with stoichiometric quantity of air. The adiabatic reaction temperature obtained is T₁. A second experiment is also conducted under identical conditions but with 25 percent excess air. The adiabatic reaction temperature obtained in the second case is T₂. Which of the following is true?

Options:

$$T_1 \leq T_2$$

$$_{2}$$
 T_{1} $< T_{2}$

$$T_1 = T_2$$

$$T_1 > T_2$$

Question Number : 200 Question Id : 67809417223 Display Question Number : Yes Single Line Question Option : No Option Orientation : Vertical

Spring-dashpot models are used to describe _____

- 1. laminar flow
- viscoelastic behavior
- 3. potential flow
- 4. molecular motion