Polymers

Classification: Natural and synthetic methods of the Polymerisation

- A high molecular weight molecule built from a large number of simple 1. molecules is called a 1) Monomer 2) Isomer 3) Polymer 4) Tautomer 2. A high molecular weight molecule which does not contain repeating structural units is called a 2) Macromolecule 1) Polymer 4) None of the above 3) Both 1 & 2 The simple molecules from which a polymer is made are called 3. 1) Monomers 2) Metamers 4) Enantiomers 3) Rotamers Which of the following is not a biopolymer? 4. 2) Nucleic acids 1) Proteins 4) Neoprene 3) Cellulose Which of the following is a synthetic polymer? 5. 1) Starch 2) Silk 3) Protein 4) Polystyrene 6. Homopolymers are made from 1) Only one type of monomers 2) Two different types of monomers 3) Three different types of monomers
 - 4) Several different types of monomers

7. Amongst the following, a homopolymer is 1) PMMA 2) Bakelite 3) Glyptal 4) Dacron 8. Which of the following is copolymer? 1) Buna-S 2) PAN ç_O, 4) PTFE 3) Polythene 9. Which of the following is a linear polymer? 1) Nylon 2) Bakelite 4) Melamine-formaldehyde polymer 3) Alkyd resin 10. Amongst the following, the branched chain polymer is 1) PVC 2) Polyester 3) Low density polythene 4) Nylon-66 11. A copolymer of acrylonitrile and 1, 3-butadiene is called 2) Polystyrene 1) Buna-N 4) Buna-S 3) Neoprene 12. Which of the following is wrong? 1) PMMA is called Plexiglas 2) PTEE is called Teflon 3) SBR is natural Rubber 4) LDPE is called low density Polythene 13. Which of the following statement/s is/are correct? 1) Vinyon is a copolymer of vinyl chloride and vinyl acetate. 2) Saran is a copolymer of isobutylene and isoprene. 3) Butyl rubber in a copolymer of isobutylene and isoprene 4) All are correct.

14. Mark the correct statement about thiokol rubber.

- 1) It is a synthetic polysulphide rubber.
- 2) It is obtained by condensation chloride with sodium tetrasulphide.
- 3) It is resistant to oils and abrasion.
- 4) All are correct.

15. Which of the following pairs is not correctly matched?

- 1) Terylene-condensation polymer of terephthalic acid and ethylene glycol
- 2) Teflon polymer of phenol and formaldehyde
- 3) Perspex-A homopolymer of methyl methacrylate
- 4) Synthetic rubber-A copolymer of butadiene and styrene
- 16. Which of the following an addition (chain growth) polymer?
 - 1) Nylon-66
- 2) Polyester
- 3) PVC 4) Glyptal

17. Which of the following is not an addition polymer?

- 1) Polystyrene 2) PVC
- 3) Polypropylene

4) Nylon

18. An example of addition copolymer is

- 1) Polythene 2) Butyl rubber
- 3) Neoprene 4) Natural rubber
- **19.** Which of the following is an addition homopolymer?
 - 1) Polythene 2) Teflon
 - 3) PVC 4) All the above
- 20. Which of the following sets contain only addition homopolymers?
 - 1) Polythene, Natural Rubber, Cellulose
 - 2) Starch, Nylon, Polyester

- 3) Teflon, Bakelite, Orlon
- 4) Neoprene, PVC, Polythene

21. Which of the following is not a condensation (step growth) polymer?

1) Melamine-formaldehyde resin 2) Bakelite 3) Polythene

4) Polyester

22. An example of a condensation homopolymer is

- 1) Bakelite
- 2) Melamine-formaldehyde resin
- 3) Alkyl resin
- 4) Perlon or Nylon-6

23. A polymer formed by coordination polymerization is

- 1) Low density polythene
- 2) High density polythene
- 3) Nylon-6
- 4) Dacron

24. Low density polythene is prepared by

- 1) Free radical polymerization
- 2) Cationic polymerization
- 3) Anionic polymerization
- 4) Ziegler-Natta polymerization

25. The best way to prepare polyisobutylene is

- 1) Coordination polymerization
- 2) Free radical polymerization
- 3) Cationic polymerization
- 4) Anionic polymerization

26. Natural rubber is

1) Polyvinyl chloride

2) cis-Polyisoprene

	3) trans-Polyisoprene	4) Polychloroprene					
27.	Gutta percha is						
	1) trans-Polyisoprene	2) A synthetic polymer					
	3) A very hard material	4) All statements are correct					
28.	Natural silk is a						
	1) Polypeptide	2) Polysaccharide					
	3) Polychloroprene	4) Polyacrylonitrile					
29.	Artificial silk is a						
	1) Polypeptide	2) Polysaccharide					
	3) Polychloroprene	4) Polyacrylonitrile					
30.	Which of the following is not a polyamid	e?					
	1) Wool	2) Leather					
	3) Nylon	4) Natural rubber					
31.	. Among the following polymers, the strongest intermolecular forces of attraction						
	are present in						
	1) Elastomers	2) Fibres					
	3) Thermoplastic	4) Thermosetting polymers					
32.	Among the following, the weakest inter-	particle forces of attraction are present					
	in S						
	1) Thermosetting polymers	2) Thermoplastic polymers					
	3) Fibres	4) Elastomers					
33.	Thermoplastics are						
	1) Linear polymers						
	2) Soften or melt on heating						
	3) Molten polymer can be moulded in desir	ed shape					
	4) All are correct						

34. Thermosetting polymers are

- 1) Cross-lined polymers
- 2) Do not melt or soften on heating
- 3) Cross-linking occurs during heating when it hardens irreversibly
- 4) All are correct

35. Which of the following is not a thermosetting polymer?

- 1) Alkyl resin 2) Bakelite
- 3) Melmac 4) SBR

36. Which are true for elastomers?

- 1) They posses electricity.
- 2) These possess weak intermolecular forces of attraction between polymer chains.
- 3) Vulcanization rubber is an example of elastomers.
- 4) All are correct.

37. Which of the following can be re-melted time and again without producing any change?

- 1) Thermosetting polymers
- 2) Thermoplastic polymers
- 3) Bakelite
- 4) Melamine-formaldehyde polymer

38. The tensile strength, elasticity and resistance to abrasion can be increased by a process called

- 1) Diazotization 2) Vulcanization
- 3) Isomerization 4) Polymerization

39. The process of vulcanization was introduced by

1) Charles Goodyear 2) Kolbe

3) Wohler

4) Zeiger

40. In vulcanization of rubber

- 1) Sulphur reacts to form a new compound
- 2) Sulphur cross-links are introduced
- 3) Sulphur forms a very thin protective layer over rubber
- 4) All statements are correct

41. Vulcanized rubber resists

- 1) Wear and Tear due to Friction
- 2) Cryogenic Temperature
- 3) High Temperature
- 4) Action of acids

42. The polymer contained by condensation of sebacic acid and hexamethylenediamine is called

- 1) Nylon 6, 6

2) Nylon 6
 4) Dacron

3) Nylon 6, 10

43. The linear chains in nylon are held together by

- 1) H-bonds 2) Covalent bonds
- 3) Ionic bonds (4) Van der waals forces

44. The monomer of PVC is

- 1) Ethylene2) Vinyl Cyanide
- 3) Vinyl chloride 4) Chloroprene

45. The repeating structural unit in neoprene is

- 1) Chloroprene 2) Chloropicrin
- 3) Chloroethene 4) ethylene

46. Chloroprene is obtained by addition of HCl to

1) Acetylene 2) Vinylacetylene

3) Divinylacetylene 4) Phenylacetylene 47. The monomer used for the manufacture of PVC is obtained by the addition of 1) HCl to acetylene in presence of Hg^{2+} salts 2) Cl_2 to acetyelene 3) HCl to ethylene 4) Cl₂ to ethylene 48. To make PVC a flexible plastic, the additive used is called 1) Filler 2) Antioxidant 3) Stabilizer 4) Plasticizer 49. Which of the following cannot be used as a plasticizer? 2) Di-n-butylphthalate 1) Diethyl phthalate 3) Di-n-octylphthalate 4) Tricresyl phosphate 50. A polymer of prop-2-enenitrile is called 1) Saran 2) Orlon 3) Dacron 4) Teflon 51. The polymer melmac is obtained by 1) Addition polymerization of melamine and formaldehyde 2) Condensation of polymerization of melamine and formaldehyde 3) Coordination polymerization of melamine 4) Free-radical polymerization of tetrafluroethylene 52. The chemical name for melamine is 1) 1, 3, 5-Triamino-2, 4, 6-triazine 2) 2, 4, 6-Triamino-1, 3, 5-triazine 3) 2-amino-1, 3, 5-triazine 4) 2, 4-Diamino-1, 3, 5-triazine 53. Starch is the condensation polymer of 1) α-D-Glucose 2) β -D- Glucose

	3) β-D- Fructose	4) α-D -Fruc	tose						
54.	Repeating disaccharide unit of starch is								
	1) Lactose	2) Sucrose							
	3) Maltose	4) Cellobios	e						
55.	Cellulose is a conden	sation polymer of							
	1) β-D-Galactose	2) α-D-Gluc	ose						
	3) β-D-Glucose	4) α-D-Galae	ctose						
56.	The repeating disacc	haride unit of cellul	ose is						
	1) Cellobiose	2) Maltose	3) Lactose 4) Sucrose						
57.	A polymer which has	s better light transm	nission properties than even glass is						
	1) Perspex		2) Bakelite						
	3) Buna-S	>	4) Poly (ethyl acrylate)						
58.	The polymer used in	manufacture of ele	ectrical goods such as switches, plugs etc						
	is								
	1) Polythene	C	2) Bakelite						
	3) Melamine formalde	hyde resin	4) Neoprene						
59.	The polymer used for	r coating electrical	wires, cables etc is						
	1) Natural rubber		2) Neoprene						
	3) Nitrile rubber		4) PVC						
60.	A polymer which is c	commonly used as a	packaging material is						
	1) Polythene		2) Polypropylene						
	3) PVC		4) Bakelite						
61.	A synthetic rubber	which is resistant t	to the action of oils, gasoline and other						
	solvents is								
	1) Buna-S		2) Polyisoprene						

- 3) Neoprene 4) Polystyrene
- 62. The monomer unit of silicone-a water repellant, heat and acid resistant polymer is
 - 1) Si 2) SiO₂ 3) R₂SiO 4) R₄Si
- 63. If N₁, N₂, N₃.....N_i are the number of molecules with molecular masses M₁, M₂
 M₃ M_i respectively, then the weight average molecular mass (M_w) is expressed as

1)
$$\frac{\sum N_i M_i^2}{\sum N_i M_i}$$

2) $\frac{\sum N_i M_i}{\sum N_i}$
3) $\frac{\sum M_i^2}{\sum N_i}$
4) $\frac{\sum N_i M_i}{\sum M_i}$

- 64. If N₁, N₂, N₃.....N_i are the number of molecules with molecular masses M₁, M₂ M₃ M_i respectively, then the number average molecular mass $(\overline{M_n})$ is expressed as
 - 1) $\frac{\Sigma N_i M_i^2}{\Sigma N_i M_i}$ 2) $\frac{\Sigma N_i M_i}{\Sigma N_i}$ 3) $\frac{\Sigma M_i^2}{\Sigma N_i}$ 4) $\frac{\Sigma N_i M_i}{\Sigma M_i}$
- 65. Number average molecular mass $(\overline{\mathbf{M}_{n}})$ and weight average molecular mass

$$(\overline{\mathbf{M}_{w}}) \text{ of synthetic polymers are related as}$$

$$1) \overline{M_{n}} < \overline{M_{w}}$$

$$2) \overline{M_{n}} > \overline{M_{w}}$$

$$3) \overline{M_{n}} = \overline{M_{w}}$$

$$4) \overline{M_{n}} = \sqrt{\overline{M_{w}}}$$

- 66. The abbreviation PDI refers to
 - 1) Name of the polymer

4

2) Polydispersity index

- 3) Planck's disposal index
- 4) Poly diagonal index

67. PDI for natural polymers is generally close to

1)	Zero		2)	100		3) 1		4) 10	
		K	ey					G	nc
		Level	- I				J	C .	
1) 3	2) 2	3) 1	4) 4	5) 4	6) 1	. 6	3,		
7) 1	8) 1	9) 1	10) 3	11) 1	12) 3				
13) 4	14) 4	15) 2	16) 3	17) 4	18) 2				
19) 4	20) 4	21) 3	22) 4	23) 2	24) 1				
25) 3	26) 2	27) 1	28) 1	29) 2	30) 4				
31) 2	32) 4	33) 4	34) 4	35) 4	36) 4				
37) 2	38) 2	39) 1	40) 2	41) 1	42) 3				
43) 1	44) 3	45) 1	46) 2	47) 1	48) 4				

49) 1 50) 2 51) 2 52) 2 53) 1 54) 3

55) 3 56) 1 57) 1 58) 2 59) 2 60) 1

61) 3 62) 3 63) 1 64) 2 65) 1 66) 2

67) 3

SOME IMPORTANT POLYMERS NATURAL AND SYNTHETIC LIKE POLYMERS LIKE POLYSTERS, BAKELITE, RUBBER, BIODEGRABLE AND NON-BIODEGRABLE POLYMERS

1.	Polymer obtained by condensation polymerization is										
	1) Polythene	2) Teflon									
	3) Phenol-formaldehyde	4) Nitrile ru	lbber								
2.	Which is an example of thermosettin	ng polymer?									
	1) Polythene	2) PVC									
	3) Neoprene	4) Bakelite									
3.	Which of the following fibres is made	e of polyamides	?								
	1) Dacron 2) Orlon 3)) Nylon	4) Rayon								
4.	Which one of the following can b	oe used as a r	nonomer in polymerization								
	reaction?										
	1) CH ₃ CH ₂ Cl	2) CH ₃ CH ₂	2OH								
	2) C6H6	4) C3H6									

5. The catalyst used in the manufacture of polyethylene by Ziegler method is

	1) Titanium tetrachloride and triphenyl aluminium								
	2) Titanium tetrachloride and triethyl aluminium								
	3) Titanium dioxide								
	4) Titanium isopropoxide								
6.	Bakelite is obtained from phenol by rea	cting it with							
	1) Acetaldehyde	2) Acetal							
	3) Formaldehyde	4) Chlorobenzene							
7.	Synthetic human hair wigs are made	from a copolymer of vinyl chloride and							
	acrylonitrile and is called								
	1) PVC	2) Polyacrylonitrile							
	3) Cellulose	4) Dynel							
8.	An example of biopolymer is								
	1) Teflon	2) Neoprene							
	3) Nylon-66	4) DNA							
9.	Synthetic polymer prepared by using ca	prolactam is known as							
	1) Terylene	2) Teflon							
	3) Nylon-6	4) Neoprene							
10.	The turbidity of polymer solution measured	ıres							
	1) The light scattered by solution								
	2) The light absorbed by a solution								
	3) The light transmitted by a solution								
	4) None of the above								
11.	Plexiglas (PMMA) is a polymer of								
	1) Acrylic acid	2) Methyl acrylate							
	3) Methyl methacrylate	4) Adipic acid							

- **13.** The monomeric unit of orlon molecule is
 - 1) CH₂=CH-Cl 2) CH₃COO-CH=CH₂
 - 3) CH₂=CH-CN 4) C₆H₅-CH=CH₂

14. Which of the following is not an example of addition polymer?

- 1) Polystyrene 2) Nylon
- 3) PVC 4) Polypropylene

15. Teflon is a polymer of monomer

- 1) Difluoroethylene 2) Monofluoroethylene
- 3) Tetrafluoroethylene 4) Trifluoroethylene
- 16. Formation of polyethylene from calcium carbide takes place as follows
 - $CaC_2 + 2H_2O \rightarrow Ca(OH)_2 + C_2H_2$
 - $C_2H_2 + H_2 \rightarrow C_2H_4$

 $nC_2H_2 \rightarrow (-CH_2 - CH_2 -)_n$

The amount of polyethylene obtained from 64.1 kg of CaC₂ is

1) 7 $l_{r_{\alpha}}$	$(2) 14 1_{20}$	3) 21 kg	$(1) 20 1_{rac}$
1) 7 kg	2) 14 kg	• 5) ZI K9	4) 28 kg
-) ·0		-)0	-) =8

17. Which one of the following is used to make 'nonstick' cookware?

1) PVC2) Polystyrene3) Poly (ethylene terephthalate)4) Polytetrafluoroethylene

18. Ebonite is

- 1) Natural rubber
- 3) Highly vulcanized rubber

19. Orlon is a polymer of

- 1) Styrene
- 3) Vinyl chloride

- 2) Synthetic rubber
- 4) Polypropylene
- 2) Tetrafluoroethylene
- 4) Acrylonitrile

20. Which of the following polymers do not involve cross linkages? 1) Melmac 2) Bakelite 4) Vulcanised rubber 3) Polythene 21. Bakelite is prepared by the reaction between 1) Phenol and formaldehyde J.Con 2) Ethylene glycol and dimethyl phthalate 3) Urea and formaldehyde 4) Tetramethylene glycol and hexamethylenediamine 22. Polymer which has amide linkage is 4) Bakelite 3) Teflon 1) Nylon-6, 6 2) Terylene 23. Glyptal polymer is obtained from glycerol by reacting with 2) Phthalic acid 1) Malonic acid 4) Acetic acid 3) Maleic acid 24. The monomer unit of polyvinyl chloride has the formula 1) $CH_3 - CH_2Cl$ 2) $CH_2 = CH_2$ 3) CHCl = CHCl 4) $CH_2 = CHCl$ 25. Which compound/set of compounds is used in the manufacture of Nylon-6,6? 1) $HOOC(CH_2)_4 COOH + H_2N(CH_2)_6 NH_2$ 2) $CH_3 = CH - C(CH_3) = CH_2$ 3) $CH_2 = CH_2$ 4) HOOC $COOH + HOCH_2 - CH_2OH$ 26. P.V.C. is formed by polymerization of 1) 1-chloroethene 2) Ethene 3) Propene 4) 1-Chloropropane

27.	Dimethyl phthalate and ethylene glycol react to form							
	1) Nylon-6	2) Nylon-66						
	3) Dacron	4) Neoprene						
28.	Which of the following contains isoprene	units?						
	1) Natural rubber	2) Nylon-66						
	3) Polyethylene	4) Dacron						
29.	Which is not a macromolecule?							
	1) DNA	2) Starch						
	3) Palmitate	4) Insulin						
30.	Which one of the following is not an exam	nple of chain growth polymer?						
1) Neoprene 2) Buna-S								
	3) PMMA	4) Glyptal						
31.	Ziegler-Natta catalyst is							
	1) $Pd + BaSo_4$							
	2) HC l + ZnC l_2							
	3) TiC l_4 + A $l(C_2H_5)_3$							
	4) LiA <i>l</i> H ₄							
32.	Nylon-66 is made by using							
	1) Phenol	2) Benzaldehyde						
	3) Adipic acid	4) Succinic acid						
33.	Natural rubber is a polymer of							
	1) Butadiene	2) Ethyne						
	3) Styrene	4) Isoprene						
34.	Terylene is a condensation polymer of et	hylene glycol and						
	1) Benzoic acid	2) Phthalic acid						

3) Salicylic acid

4) Terephthalic acid

35. Which is not true about polymers?

- 1) Polymers do not carry any charge.
- 2) Polymers have high viscosity.
- 3) Polymers scatter light.
- 4) Polymers have low molecular weight.

36. On the basis of mode of formation, polymers can be classified?

- 1) As addition polymers only
- 2) As condensation polymers only
- 3) As copolymers
- 4) Both as addition and condensation polymers
- 37. The process of involving heating of rubber with sulphur is called

1) Galvanization	2) Vulcanization
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- 3) Bessemerisation 4) Sulphonation
- 38. Terylene is made by polymerization of terephthalic acid with
 - 1) Ethylene glycol
 2) Phenol
 3) Ethanol
 4) Catechol
- 39. Teflon, styrene and neoprene are all
 - 1) Copolymers2) Condensation polymers3) Homopolymers4) Monomers
- 40. Inter-particle forces present in Nylon-6,6 are
 - 1) Van der Waal's2) Hydrogen bonding
 - 3) Dipole-dipole interactions 4) None of these
- 41. $F_2C=CF_2$ is a monomer of

1) Teflon	2) Glyptal
3) Nylon-6	4) Buna-5

42. Soft drinks and baby feeding bottles are generally made up of 1) Polyester 2) Polyurethane 4) Polyamide 3) Polyurea 43. Polymer used in bullet proof glass is 1) PMMA 2) Lenan 3) Nomex 4) Kevlar у. С С 44. Which of the following is a constituent of nylon? 1) Adipic acid 2) Styrene 3) Teflon 4) None of these 45. Caprolactam polymerises to give 1) Terylene 2) Teflon 3) Glyptal 4) Nylon-6 46. Polyvinyl alcohol can be prepared by 1) Polymerization of vinyl alcohol 2) Alkaline hydrolysis of polyvinyl acetate 3) Polymerization of acetylene 4) Reaction of acetylene with H₂SO₄ in presence of HgSO₄ 47. A condensation polymer among the following is 2) PVC 1) Dacron 3) Polystyrene 4) Teflon 48. The catalyst used for the polymerization of olefins is 1) Zieglar-Natta catalyst 2) Wilkinson's catalyst 3) Pd-catalyst 4) Zeise's salt complex 49. Which of the following used in paints? 1) Terylene 2) Nylon 3) Glyptal 4) Chloroprene

50.	Polymer formation from monomers starts by											
	1) Condensation reaction between monomers											
	2) Coordination reaction between monomers											
	3) Conversion of monomer to monomer ions by protons											
	4) Hydrolysis of monomers											
51.	Which of	the follow	ing is a po	olyan	nide molecule?							
	1) Teryler	le			2) Rayon							
	3) Nylon-	б			4) Polystyrene							
52.	Nylon thr	eads are n	nade of									
	1) Polyeth	ylene poly	mer		2) Polyvinyl polymer							
	3) Polyest	er polymer			4) Polyamide polymer							
53.	Rubber is	s a polyme	r of									
	1) Pyrene				2) Isoprene							
	3) Urea			4) Ethylene								
54.	Which of	the follow	ing is curi	rently	y used as a tyre cord?							
	1) Teryler	ie	C		2) Polyethylene							
	3) Polypro	pylene	N-		4) Nylon-6							
55.	List - I		0	List - II								
	A) Natur	al Polymer		1) PVC								
	B) Synth	etic Polym	er	2) Nylon - 6, 6								
	C) Conde	ensation po	olymer	3) Silk								
	D) Addit	ion polyme	er	4) Polyethylene								
		Α	B	С	D							
	1)	2	3	4	1							
	2)	3	2	1	4							
	3)	3	4	2	1							

	4)	3	2	4	- 1						
56.	List - I			List	- II						
	A) Rayor	1		1) Cationi	c Polyn	nerisati	on			
	B) Glass			2	2) Organic	e Polym	ers				
	C) Polyst	yrene		3	8) Inorgan	ic poly	mers				^
	D) Poly i	soprene		4) Anionic	e polym	erisatic	n			
		Α	В	(C D				(-Q	•
	1)	2	3	4	- 1						
	2)	3	2	4	1						
	3)	2	1	3	8 4			$\langle \mathbf{V} \rangle$			
	4)	2	3	1	4		0				
					5	5					
				X	Key	7					
1)	3 2) 4	3) 3	4) 4	5) 2	6) 3	7) 4	8) 4	9) 3	10) 1	11) 3	12) 3
13)	3 14) 2	15) 3	16) 4	17) 4	18) 3	19) 4	20) 3	21) 1	22) 1	23) 2	24) 4
25)	1 26) 1	27) 3	28) 1	29) 3	30) 4	31) 3	32) 3	33) 4	34) 4	35) 4	36) 4
	2 38) 1										
49)	3 50) 1	51) 3	52) 4	53) 1	54) 2	55) 2	56) 4				