# www.sakshieducation.com <a href="mailto:ksakshieducation.com"><u>Expansion of Liquids</u></a>

- 1. Linear and areal expansions have no significance for a liquid, since shape of liquid depends on shape of vessel.
- **2.** Liquids do not possess any definite shape and require a container to hold them. Hence only cubical expansion is considered.
- 3. For liquids there are two types of cubical expansion
  - i) Coefficient of apparent expansion  $(\gamma_a)$
  - ii) Coefficient of real or absolute expansion  $(\gamma_r)$
- **4.** Coefficient of apparent expansion of a liquid is the ratio of the apparent increase in volume per 1°C rise of temperature to its initial volume.

$$\gamma_a = \frac{\text{apparent increase in volume}}{\text{original volume x rise in temperature}}$$

Unit of 
$$\gamma_a$$
:  ${}^{\circ}C^1$ .

5. Coefficient of real expansion is the ratio between real increase in volume per 1°C rise of temperature and the original volume of the liquid.

$$\gamma_{\rm r} = {
m real\,increase\,in\,volume \over 
m original\,volume\,\,x\,rise\,in\,temperature} = {V_2 - V_1 \over V_1(t_2 - t_1)}$$

Unit of 
$$\gamma_r : {}^{\circ}C^1$$
.

- 6.  $\gamma_r = \gamma_a + \gamma_{vessel} = \gamma_a + 3\alpha$ .
- 7. If  $\gamma_v = +ve$  and  $\gamma_v < \gamma_v$ ,  $\gamma_a = ve$ , the level decreases continuously when heated.
- 8. If  $\gamma_v = +ve$  and  $\gamma_r = \gamma_v$ ;  $\gamma_a = 0$ , the level will not change when heated.
- 9. If  $\gamma_v = +ve$  and  $\gamma_r > \gamma_v$ ;  $\gamma_a = +ve$ , the level first falls and then rise when heated.
- 10. If  $\gamma_v$ =0;  $\gamma_r$ = $\gamma_a$ , the level will increase continuously when heated.
- 11. If  $\gamma_v = ve$ ,  $\gamma_a > \gamma_r$ , the level will increase continuously when heated.
- 12. The real expansion of a liquid does not depend upon the temperature of the container.
- **13.** The apparent expansion of liquid depends on a) initial volume or liquid, b) rise in temperature c) nature of liquid and d) nature of container.
- **14.** The fraction of the volume of a glass flask that must be filled with mercury so that the volume of the empty space left may be the same at all temperatures is 1/7.

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- **15.** If the same liquid is heated in two different vessels x and y then  $(r_a)_x + 3\alpha_t = (r_a)_y + 3\alpha_y$
- **16.** In determining  $\gamma_r$  by Dulong and Pettit's method, if  $h_0$  and  $h_t$  are heights of liquid is the two limbs of a U tube maintained at 0°C and t°C  $\gamma_r = \frac{h_t h_0}{h_t t}$  /°C
- 17. In determining  $\gamma_a$  by specific gravity bottle method.

Weight of empty bottle =  $w_1g$ 

Weight of (bottle + liquid) =  $w_2g$ 

Weight of (bottle + remaining liquid) =  $w_3g$ 

$$\gamma_{app} = \frac{(w_3 - w_2)}{(w_3 - w_1)(t_2 - t_1)} / ^{\circ} C$$

$$\gamma_{app} = \frac{Massof \ the \ liquid \ exp \ elled}{Massof \ the \ remaining \ liquid \times Temp.diff}.$$

- **18.** If m<sub>1</sub> is mass of liquid and m<sub>2</sub> is mass of remaining liquid, then  $\gamma_a = \frac{m_1 m_2}{m_2(t_2 t_1)}$
- **19.** A specific gravity bottle contains m g of liquid of apparent expansion  $\gamma$  at 0°C. if it is heated through t°C, mass of expelled liquid is  $x = \frac{\gamma mt}{1 + \gamma t}$
- **20.** When a solid substance is immersed in a liquid at 0°C, the loss of weight is  $w_0$ . Now liquid is heated to t°C, loss of weight in the substance is  $w_t$ . Then  $w_t = w_0 \left[ 1 + (\gamma_s \gamma_R)t \right]$
- **21.** A solid of negligible volume expansion floats in a liquid. At  $t_1$ °C &  $t_2$ °C. the fractions  $f_1$ ,  $f_2$  of volumes of the solid remains submerged in the liquid. The coefficient of volume expansion of liquid is  $\gamma_R = \frac{a_2 a_1}{a_1 t_2 a_2 t_1}$
- 22. To keep the volume of empty space in a vessel (volume  $v_g$ ) constant at all temperatures by pouring certain amount of a liquid of volume  $v_l$ , the condition is  $v_l\gamma_l=v_g\gamma_g$  where  $\gamma_l=$ coefficient of cubical expansion of liquid and  $\gamma_g=$ coefficient of cubical expansion of vessel.
- **23.** The density of a liquid usually decreases when heated. If d<sub>1</sub> and d<sub>2</sub> are the densities of a liquid at 0°C and t°C respectively, then

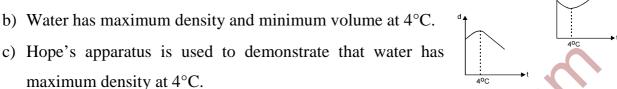
$$d_t = \frac{d_o}{1 + \gamma_r t}$$
;  $d_t = d_o(\tilde{1}\gamma_r t)$ ;

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$$\gamma_r = \frac{d_1 - d_2}{d_1 t_2 - d_2 t_1} / {}^{o} C$$

#### 24. Anomalous Expansion of Water

a) When water at 0°C is heated, its volume decreases up to 4°C and from 4°C its volume increases with the increase of temperature. This peculiar behavior of water is called anomalous expansion of water.



- d) Dilatometer is used to prove anomalous expansion of water.
- e) Aquatic animals are surviving in cold countries due to the anomalous expansion of water.
- f) During winter, in cold countries, even if the temperature falls far below 0°C, the water in the frozen lakes or seas at the bottom remains at 4°C.
- g) When water freezes, it expands and consequently water pipes burst in winter.
- h) When water at 4°C is filled to the brim of a beaker, then it over flows when it is either cooled or heated.
- i) Water has positive coefficient of expansion above 4°C and negative coefficient below 4°C.
- j) At 4°C the coefficient of expansion of water is zero.