ALDEHYDES, KETONES and CARBOXYLIC ACIDS

1.	Which one of the following gives yellow precipitate with iodine and NaOH solution ? (EAMCET-2010)
	1) $CH_3 - CHO$ 2) $C_6H_5COC_6H_5$ 3) HCHO 4) CH_3OH
2.	What are X and Y in the following reaction sequence ?
	$C_2H_5OH \xrightarrow{Cl_2} X \xrightarrow{Cl_2} Y $ (EAMCET-2009)
	1) C ₂ H ₅ Cl, CH ₃ CHO 2) CH ₃ CHO, CH ₃ CO ₂ H
	3) CH ₃ CHO, CCl ₃ CHO 4) C ₂ H ₅ Cl, CCl ₃ CHO
3.	Acetone on addition to methyl magnesium bromide forms a complex, which on decomposition with acid gives X and Mg(OH)Br. Which one of the following is X ?
	1) CH ₃ OH 2) (CH ₃) ₃ COH (EAMCET-2008)
	3) (CH ₃) ₂ CHOH 4) CH ₃ CH ₂ OH
4.	Identify A and B in the following reaction
	$CH_3 - CH_3 \xleftarrow{B} CH_3 COOH \xrightarrow{A} CH_3 CH_2 OH$ (EAMCET-2008)
	A B
	1) HI + red P LiAlH ₄
	2) Ni/Δ LiAlH ₄
	3) LiAlH ₄ HI + red P
	4) Pb - BaSO ₄ $Zn + HCl$
5.	$CH_{3}COOH \xrightarrow{LiAlH_{4}} A \qquad A + CH_{3}COOH \xrightarrow{H_{2}O^{+}} B + H_{2}O$ 'A' and 'B' respectively, are In the above reactions (EAMCET-2007)
	1) $CH_3COOC_2H_5$, C_2H_5OH 2) CH_3CHO , C_2H_5OH (EAWCE 1-2007)
	3) C ₂ H ₅ OH, CH ₃ CHO 4) C ₂ H ₅ OH, CH ₃ COOC ₂ H ₅
6.	An organic compound X on treatment with pyridium dichromate in dichloromethane gives compound Y. Compund Y reacts with and alkali to form idodoform. Th compound X is (EAMCET-2007)
	$\begin{array}{c} C_2H_5OH \\ 2) CH_3CHO \end{array}$
	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $
7.	What reagent is used in Rosenmund's reduction?(EAMCET-2006)
	$_{1)} H_2 / Pd - BaSO_4 \qquad \qquad _{2)} LiAlH_4$
	3) $NH_2 - NH_2 / KOH / CH_2OH - CH_2OH$ 4) $Zn - Hg / HCl$

8. Which of the products formed when acetone is reacted with barium hydroxide solution?

```
(EAMCET-2006)
```

$$CH_{3} - C - CH_{2} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{2} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{2} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{3} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{3} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{3} - CH_{3} - CH_{3}$$

$$CH_{3} - C - CH_{3} - CH_{3} - CH_{3}$$

9. Which of the following reagents converts both acetaldehyde and acetone to alkanes?

(EAMCET 2006)

- $\begin{array}{ccc} & & & \\ 1) & \text{Ni} / \text{H}_2 & & \\ 2) & \text{LiAlH}_4 \\ \hline \\ 3) & \text{I}_2 / \text{NaOH} & & \\ 4) & \text{Zn} \text{Hg} / \text{Con.HCl} \end{array}$
- 10. 3-Hydroxybutanal is formed when(X) reacts with (Y) in dilute (Z) solution. What are X, Y and Z ? (EAMCET- 2005)

	X	Y	Z
1.	CH ₃ CHO	(CH ₃) ₂ CO,	NaOH
2	CH ₃ CHO	CH ₃ CHO	NaCl
3.	(CH ₃) ₂ CO,	(CH ₃) ₂ CO,	HC1
4.	СН3СНО,	CH ₃ CHO,	NaOH

11. A compound 'X' undergoes reaction with LiAlH_4 to yield 'Y'. When vapours of 'Y' are passed over freshly reduced copper at 300° C. 'X' is formed what is 'Y'? (EAMCET-2005)

 $\begin{array}{ccc} & & & \\ 1) & CH_3COCH_3 & & \\ 3) & CH_3CH_2OH & & \\ \end{array} \begin{array}{c} & & \\ 2) & CH_3CHO & \\ & & \\ 4) & CH_3-O-CH_3 \end{array}$

12. Which of the following reagents can form a phenyl hydrazone with alkanone ? (EAMCET - 2004)

- $\overset{\oplus}{N}H_{3}OHC\overset{\oplus}{l} 2) PhNHNH_{2}$ $3) NH_{2}NHCONH_{2}$ 4) HCN
- 13. When ethyl alcohol is pased over red hot copper at the formula of the product formed is

(EAMCET 2003)

www.sakshieducation.com

14. Acetaldehyde forms white crystalline precipitate on mixing with asolution of (EAMCET 2002) 2) Alcoholic 1) Acidic Zn, Hg 3) Saturated, aqueous $NaHSO_3$ 4) Aqueous NaCl 15. The chemicals used for preparing acetophenone are (EAMCET 2002) $A) C_6 H_6$ CH₃COOCH₃ B) C) CH₃COCl D) Anhydrous 1) A,B,C 2) B,C,D 3) A,C,D 4) A,B,D 16. Dry distillation of a mixture of calcium acetate and calcium formate forms. (EAMCET2001) 1) Methanol 2) Ethanal 3) Ethanol 4) Acetone 17. Which of the following converts acetone to acetone oxime ? (EAMCET2000) 1) H_2N-NH_2 2) 2, 4-DNP 3) C₆H₅NHNH₂ 4) NH₂OH Identify A and B in the following reaction $CH_3 - CH_3 \leftarrow B - CH_3COOH - A \rightarrow CH_3CH_2OH$ 18. A В (EAMCET-2008) 1) HI + red PLiAlH₄ 2) Ni/Δ LiAlH₄ HI + red P 3) LiAlH₄ 4) Pb - BaSO₄ Zn + HCl $\xrightarrow{H_2O^+} B + H_2O$ $CH_3COOH \xrightarrow{LiAlH_4} A$ $A + CH_2COOH -$ In the above reactions 'A' and 19. 'B' respectively, are (EAMCET-2007) 1) CH₃COOC₂H₅, C₂H₅OH 2) CH₃CHO, C₂H₅OH 3) C₂H₅OH, CH₃CHO 4) C₂H₅OH, CH₃COOC₂H₅ 20. Which of the following is a pair of functional isomers ? **(EAMCET2005)** 1) CH3COCH3, CH3CHO 2) C₂H₅CO₂H, CH₃CO₂CH₃ 3) C₂H₅CO₂H, CH₃CO₂C₂H₅ 4) CH₃CO₂H, CH₃CHO Acid hydrolysis of X yields two different organic compounds. Which one of the following is X? 21. (EAMCET2003) 1) CH₃COOH 2)CH₃CONH₂ 3) CH₃COOC₂H₅ 4) (CH₃CO)₂O 22. In the reaction sequence, C₂H₅Cl +KCN.What is the molecular formula of Y? (EAMCET2003) 2) C₃H₅N 1) $C_{3}H_{6}O_{2}$ 3) C₂H₄O₂ 4) C₂H₆O 23. In the following reaction, X and Y are respectively : CH₃COOH+NH₃ \rightarrow X $\xrightarrow{\Delta}$ Y+H₂O (EAMCET2002) 1) CH₃CONH₂,CH₄ 2) CH₃COONH₄, CH₃CONH₂ 3) CH₃CONH₂, CH₃COOH 4) CH₃NH₂,CH₃CONH₂

www.sakshieducation.com

24. Aceto phenone when reacted with a base C₂H₅ONa, yields a stable compound which has the structure (AIPMT 2008)

